

- (b) Why the rate of above reaction decreases on addition of H_2O_2 [3]

UNIT-III

6. (a) Discuss nature of bonding in metal carbonyls. [5]
(b) Giving reason arrange following in order of increasing C-O bond order : [3]
 $[Ni(CO)_4]_d$, $[Co(CO)_4]_d$, $[Fe(CO)_5]_d$
(c) Give reason for the stability of square planar organometallic complexes of Ni (II), although these complexes do not obey $18e^-$ rule. [2]
7. (a) Discuss the rate of reactivity of organoberyllium compounds with N, O, S and P donors. [4]
(b) M-C bond cleavage reaction in Transition metal alkyls. [3]
(c) Why σ -bonded organometallic compounds of transition metals are less stable compared to their main group analogue. [3]

UNIT-IV

8. (a) Discuss microwave spectroscopy and give selection rule for rotational spectra of a molecule [5]

- (b) First line corresponding to $J=0$ appears at 3.84235 cm^{-1} in the rotational spectrum of $^{12}C^{16}O$ molecule (C-12 O-16). Calculate its moment of inertia and bond length. [5]

$$(h = 6.626 \times 10^{-27} \text{ erg/sec}, N = 6.022 \times 10^{23})$$

9. (a) Discuss application of IR spectra in elucidating the structure and properties of polyatomic molecules. [5]
(b) Explain why in IR spectrum of CO_2 molecule, only two absorption bands at 2349 cm^{-1} and 667 cm^{-1} respectively are observed. [5]

OR

Discuss the factors which affect the band position and intensities in I.R. region.

--- x ---

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Christian G. Meyer



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Table of Contents Chemistry Question Paper 201

1. Understanding the eBook Chemistry Question Paper 201
 - The Rise of Digital Reading Chemistry Question Paper 201
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Question Paper 201
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry Question Paper 201
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Question Paper 201
 - Personalized Recommendations

- Chemistry Question Paper 201 User Reviews and Ratings
- Chemistry Question Paper 201 and Bestseller Lists
- 5. Accessing Chemistry Question Paper 201 Free and Paid eBooks
 - Chemistry Question Paper 201 Public Domain eBooks
 - Chemistry Question Paper 201 eBook Subscription Services
 - Chemistry Question Paper 201 Budget-Friendly Options
- 6. Navigating Chemistry Question Paper 201 eBook Formats
 - ePub, PDF, MOBI, and More
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 - Chemistry Question Paper 201 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemistry Question Paper 201
 - Highlighting and Note-Taking Chemistry Question Paper 201
 - Interactive Elements Chemistry Question Paper 201
- 8. Staying Engaged with Chemistry Question Paper 201
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry Question Paper 201
- 9. Balancing eBooks and Physical Books Chemistry Question Paper 201
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry Question Paper 201
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Chemistry Question Paper 201
 - Setting Reading Goals Chemistry Question Paper 201
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chemistry Question Paper 201
 - Fact-Checking eBook Content of Chemistry Question Paper 201

- Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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