

**CHAPTER 9 REVIEW***Stoichiometry***SECTION 1****SHORT ANSWER** Answer the following questions in the space provided.

1. \_\_\_\_\_ The coefficients in a chemical equation represent the
  - (a) masses in grams of all reactants and products.
  - (b) relative number of moles of reactants and products.
  - (c) number of atoms of each element in each compound in a reaction.
  - (d) number of valence electrons involved in a reaction.
  
2. \_\_\_\_\_ Which of the following would not be studied within the topic of stoichiometry?
  - (a) the mole ratio of Al to Cl in the compound aluminum chloride
  - (b) the mass of carbon produced when a known mass of sucrose decomposes
  - (c) the number of moles of hydrogen that will react with a known quantity of oxygen
  - (d) the amount of energy required to break the ionic bonds in  $\text{CaF}_2$
  
3. \_\_\_\_\_ A balanced chemical equation allows you to determine the
  - (a) mole ratio of any two substances in the reaction.
  - (b) energy released in the reaction.
  - (c) electron configuration of all elements in the reaction.
  - (d) reaction mechanism involved in the reaction.
  
4. \_\_\_\_\_ The relative number of moles of hydrogen to moles of oxygen that react to form water represents a(n)
  - (a) reaction sequence.
  - (b) bond energy.
  - (c) mole ratio.
  - (d) element proportion.
  
5. Given the reaction represented by the following unbalanced equation:  $\text{N}_2\text{O}(g) + \text{O}_2(g) \rightarrow \text{NO}_2(g)$ 
  - a. Balance the equation.  
\_\_\_\_\_  
\_\_\_\_\_
  - b. What is the mole ratio of  $\text{NO}_2$  to  $\text{O}_2$ ?  
\_\_\_\_\_
  - c. If 20.0 mol of  $\text{NO}_2$  form, how many moles of  $\text{O}_2$  must have been consumed?  
\_\_\_\_\_
  - d. Twice as many moles of  $\text{NO}_2$  form as moles of  $\text{N}_2\text{O}$  are consumed. True or False?  
\_\_\_\_\_
  - e. Twice as many grams of  $\text{NO}_2$  form as grams of  $\text{N}_2\text{O}$  are consumed. True or False?  
\_\_\_\_\_

# Chapter 9 Review Stoichiometry Mixed Answers Chinafslore

**AN Whitehead**



**Chapter 9 Review Stoichiometry Mixed Answers Chinafslore:**

## Decoding **Chapter 9 Review Stoichiometry Mixed Answers Chinafslore**: Revealing the Captivating Potential of Verbal Expression

In a time characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its capability to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Chapter 9 Review Stoichiometry Mixed Answers Chinafslore**," a mesmerizing literary creation penned with a celebrated wordsmith, readers embark on an enlightening odyssey, unraveling the intricate significance of language and its enduring impact on our lives. In this appraisal, we shall explore the book's central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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