

Section 12.1

OBJECTIVE 1

Identify suitable Lewis acid-base adducts from the species provided:

1. BF_3
 - (a) NH_3
 - (b) H_2O
 - (c) CH_3OH
 - (d) C_2H_6
 - (e) C_2H_4
 - (f) C_2H_2
 - (g) C_2H_6
 - (h) C_2H_4
 - (i) C_2H_2
2. AlCl_3
 - (a) NH_3
 - (b) H_2O
 - (c) CH_3OH
 - (d) C_2H_6
 - (e) C_2H_4
 - (f) C_2H_2
 - (g) C_2H_6
 - (h) C_2H_4
 - (i) C_2H_2
3. FeCl_3
 - (a) NH_3
 - (b) H_2O
 - (c) CH_3OH
 - (d) C_2H_6
 - (e) C_2H_4
 - (f) C_2H_2
 - (g) C_2H_6
 - (h) C_2H_4
 - (i) C_2H_2
4. SnCl_4
 - (a) NH_3
 - (b) H_2O
 - (c) CH_3OH
 - (d) C_2H_6
 - (e) C_2H_4
 - (f) C_2H_2
 - (g) C_2H_6
 - (h) C_2H_4
 - (i) C_2H_2
5. TiCl_4
 - (a) NH_3
 - (b) H_2O
 - (c) CH_3OH
 - (d) C_2H_6
 - (e) C_2H_4
 - (f) C_2H_2
 - (g) C_2H_6
 - (h) C_2H_4
 - (i) C_2H_2

6. Write the Lewis adducts formed by the following acid-base reactions:



a. Write the equation:



b. Write the structure of $\text{BF}_3 \cdot \text{C}_2\text{H}_6$.

c. Is addition of BF_3 to C_2H_6 an exothermic reaction? Justify your answer.

d. Write the Lewis structure of $\text{BF}_3 \cdot \text{C}_2\text{H}_6$ and label the Lewis acid and Lewis base.

e. Write the Lewis structure of $\text{BF}_3 \cdot \text{C}_2\text{H}_6$ and label the Lewis acid and Lewis base.

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James R. Braun, Sherman



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