

Class 10th Objective Question

1. CaCO_3 को CaO में अपघटित करने के लिए किस प्रकार की ऊष्मा चाहिए ?
A) गर्म B) ठंडा C) गर्म D) ठंडा
2. CaCO_3 को CaO में अपघटित करने के लिए किस प्रकार की ऊष्मा चाहिए ?
A) गर्म B) ठंडा C) गर्म D) ठंडा
3. CaCO_3 को CaO में अपघटित करने के लिए किस प्रकार की ऊष्मा चाहिए ?
A) CaCO_3 B) MgCO_3 C) $\text{Ca}(\text{HCO}_3)_2$ D) None
4. H_2O को H_2 और O_2 में अपघटित करने के लिए किस प्रकार की ऊष्मा चाहिए ?
A) 1:1 B) 2:1 C) 3:1 D) 2:2
5. H_2O को H_2 और O_2 में अपघटित करने के लिए किस प्रकार की ऊष्मा चाहिए ?
A) गर्म B) ठंडा C) गर्म D) ठंडा
6. H_2O को H_2 और O_2 में अपघटित करने के लिए किस प्रकार की ऊष्मा चाहिए ?
A) गर्म B) ठंडा C) गर्म D) ठंडा
7. $2\text{Cu} + \text{O}_2 \rightarrow 2\text{CuO}$ में Cu का ऑक्सीकरण संख्या क्या है ?
A) 0 B) +1 C) +2 D) (A) और (B) दोनों
8. $\text{CuO} + \text{H}_2 \rightarrow \text{Cu} + \text{H}_2\text{O}$ में Cu का ऑक्सीकरण संख्या क्या है ?
A) 0 B) +1 C) +2 D) +3
9. $\text{CaCO}_3(\text{s}) \rightarrow \text{CaO}(\text{s}) + \text{CO}_2(\text{g})$ में CaCO_3 का ऑक्सीकरण संख्या क्या है ?
A) 0 B) +2 C) +4 D) +6
10. $\text{CaCO}_3(\text{s}) \rightarrow \text{CaO}(\text{s}) + \text{CO}_2(\text{g})$ में CO_2 का ऑक्सीकरण संख्या क्या है ?
A) 0 B) +2 C) +4 D) +6
11. $\text{CaCO}_3 + \text{CaO} + \text{CO}_2$ B) $\text{CaO} + 2\text{HCl} \rightarrow \text{CaCl}_2 + \text{H}_2\text{O}$ C) $\text{Fe} + \text{CuSO}_4 \rightarrow \text{FeSO}_4 + \text{Cu}$ D) $\text{NaOH} + \text{HCl} \rightarrow \text{NaCl} + \text{H}_2\text{O}$
12. CaCO_3 को CaO में अपघटित करने के लिए किस प्रकार की ऊष्मा चाहिए ?
A) गर्म B) ठंडा C) गर्म D) ठंडा
13. CaCO_3 को CaO में अपघटित करने के लिए किस प्रकार की ऊष्मा चाहिए ?
A) Na B) Mg C) Cu D) K
14. $\text{Na}_2\text{SO}_4(\text{aq}) + \text{BaCl}_2(\text{aq}) \rightarrow \text{BaSO}_4(\text{s}) + 2\text{NaCl}(\text{aq})$ में BaSO_4 का ऑक्सीकरण संख्या क्या है ?
A) 0 B) +2 C) +4 D) +6
15. $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$ में H_2 का ऑक्सीकरण संख्या क्या है ?
A) 0 B) +1 C) +2 D) +3
16. $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$ में O_2 का ऑक्सीकरण संख्या क्या है ?
A) 0 B) +1 C) +2 D) +3
17. $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$ में H_2 का ऑक्सीकरण संख्या क्या है ?
A) 0 B) +1 C) +2 D) +3
18. $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$ में O_2 का ऑक्सीकरण संख्या क्या है ?
A) 0 B) +1 C) +2 D) +3
19. $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$ में H_2 का ऑक्सीकरण संख्या क्या है ?
A) 0 B) +1 C) +2 D) +3
20. $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$ में O_2 का ऑक्सीकरण संख्या क्या है ?
A) 0 B) +1 C) +2 D) +3
21. $\text{ZnO} + \text{C} \rightarrow \text{Zn} + \text{CO}$ में ZnO का ऑक्सीकरण संख्या क्या है ?
A) 0 B) +2 C) +4 D) +6

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