



Classification of Crystalline Compounds

Identifying Polymers, Solids, Metals, and Liquids

Rank of Crystalline Compounds

When crystalline polymers are subjected to stress, they will undergo a phase transition and exhibit a change in their physical properties. This is because the crystalline regions of the polymer are more ordered and have a higher density than the amorphous regions. The transition from a crystalline to an amorphous state is called melting, and the transition from an amorphous to a crystalline state is called crystallization. The melting point of a polymer is the temperature at which the crystalline regions melt, and the crystallization point is the temperature at which the amorphous regions crystallize. The melting point of a polymer is a function of its molecular weight and the degree of crystallinity. The crystallization point of a polymer is a function of its molecular weight and the degree of crystallinity.

When a polymer is subjected to stress, it will undergo a phase transition and exhibit a change in its physical properties. This is because the crystalline regions of the polymer are more ordered and have a higher density than the amorphous regions. The transition from a crystalline to an amorphous state is called melting, and the transition from an amorphous to a crystalline state is called crystallization. The melting point of a polymer is the temperature at which the crystalline regions melt, and the crystallization point is the temperature at which the amorphous regions crystallize. The melting point of a polymer is a function of its molecular weight and the degree of crystallinity. The crystallization point of a polymer is a function of its molecular weight and the degree of crystallinity.

Questions: Rank of crystalline compounds

1. Which compound is not crystalline? (a) Sodium chloride (b) Sugar (c) Ice (d) Glass
2. Which compound is not crystalline? (a) Sodium chloride (b) Sugar (c) Ice (d) Glass
3. Which compound is not crystalline? (a) Sodium chloride (b) Sugar (c) Ice (d) Glass
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8. Which compound is not crystalline? (a) Sodium chloride (b) Sugar (c) Ice (d) Glass
9. Which compound is not crystalline? (a) Sodium chloride (b) Sugar (c) Ice (d) Glass
10. Which compound is not crystalline? (a) Sodium chloride (b) Sugar (c) Ice (d) Glass

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Christian Drosten



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