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**Léo Pomar**



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**Report of the Board of Education** Great Britain. Board of Education,1921    *Perry & co's monthly illustrated price current* Perry and co, ltd,1874    Report of Work of the Agricultural Experiment Station of the University of California ... California Agricultural Experiment Station,1924    *Sotheran's Price Current of Literature* ,1905    **New York School Journal** ,1899    **Bulletin** ,1892    The Farmer's Magazine Volume the Forty-Ninth January-June MDCCLXXVI January,1876    The Farmers Magazine January,1876,1876    Reports on the Botanic Station, Agricultural School, Land Settlement Scheme and of the Government Veterinary Surgeon, St. Vincent Saint Vincent. Agricultural Department,1916

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web when the reactants have more potential energy than the products have and the extra energy is released to the surroundings it is called activation energy the height of the hill on a potential energy diagram is a measure of the

endothermic the reaction above can be categorized as

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web this is like a one dimensional system whose mechanical energy  $e$  is a constant and whose potential energy with respect to zero energy at zero displacement from the spring  $s$  unstretched length  $x_0$  is  $u = \frac{1}{2} kx^2$  figure 8 5 2 8 5 2 a a glider between springs on an air track is an example of a horizontal mass spring system

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web 1 if the  $pe$  of the reactants is 430kj and the  $pe$  of the products is 190kj what is the  $\Delta h$  2 if the  $pe$  of the reactants is 430kj and the activation energy is 220 how much total energy will be needed to be added for the reaction to proceed transcribed image text potential energy diagram answer the questions using the graph

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web figure 5 5 1 a potential energy diagram shows the total potential energy of a reacting system as the reaction proceeds a in an endothermic reaction the energy of the products is greater than the energy of the reactants and  $\Delta h$  is positive b in an exothermic reaction the energy of the products is lower than the energy of the

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web answer see below explanation the rate of a chemical reaction  $a + b \rightarrow c + d$  is given in first approximation by  $v = k_v \frac{a}{b}$  where  $k_v = k e^{-E_a/RT}$  where  $k$  depends by many factors as solvent geometry type of reaction  $E_a$  is the activation energy  $R$  is the universal constant of gases  $T$  is the absolute temperature

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web aug 8 2022 the energy changes that occur during a chemical reaction can be shown in a diagram called a potential energy diagram or sometimes called a reaction progress curve a potential energy diagram shows the change in potential energy of a system as reactants are converted into products the figure below shows basic potential energy

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web key points potential energy is energy that has the potential to become another form of energy an object's potential

energy depends on its physical properties and position in a system potential energy comes in many forms such as gravitational potential energy due to an object's mass and position in a gravitational field

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web the energy of the reactants is shown on the energy diagram above on the right side of the energy diagram draw a horizontal line segment to indicate the energy of the products draw a vertical line segment to indicate  $\Delta h$  for the reaction and label it with the correct value see attached image

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web all questions are answered herein chemistry tutorial 9 02b potential energy diagrams how to draw and label pe diagrams for exothermic and endothermic reactions and what effect adding a catalyst or inhibitor has on the diagram

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time ms 0 00 1 00 2 00 3 00 4 00 5 00 6 00 7 00 8 00 9 00 10 0 potential energy kcal 0 00 10 0

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