



ADDING FRACTIONS SHEET 1 ANSWERS

$$1) \frac{1}{3} + \frac{3}{6} = \frac{2}{6} + \frac{3}{6} = \frac{5}{6}$$

$$2) \frac{1}{4} + \frac{4}{12} = \frac{3}{12} + \frac{4}{12} = \frac{7}{12}$$

$$3) \frac{1}{5} + \frac{4}{15} = \frac{3}{15} + \frac{4}{15} = \frac{7}{15}$$

$$4) \frac{1}{2} + \frac{3}{8} = \frac{4}{8} + \frac{3}{8} = \frac{7}{8}$$

$$5) \frac{2}{7} + \frac{5}{14} = \frac{4}{14} + \frac{5}{14} = \frac{9}{14}$$

$$6) \frac{1}{4} + \frac{5}{16} = \frac{4}{16} + \frac{5}{16} = \frac{9}{16}$$

$$7) \frac{3}{5} + \frac{5}{20} = \frac{12}{20} + \frac{5}{20} = \frac{17}{20}$$

$$8) \frac{1}{9} + \frac{3}{18} = \frac{2}{18} + \frac{3}{18} = \frac{5}{18}$$

$$9) \frac{3}{16} + \frac{1}{8} = \frac{3}{16} + \frac{2}{16} = \frac{5}{16}$$

$$10) \frac{1}{18} + \frac{2}{3} = \frac{1}{18} + \frac{12}{18} = \frac{13}{18}$$

$$11) \frac{3}{25} + \frac{2}{5} = \frac{3}{25} + \frac{10}{25} = \frac{13}{25}$$

$$12) \frac{8}{21} + \frac{3}{7} = \frac{8}{21} + \frac{9}{21} = \frac{17}{21}$$

$$13) \frac{4}{30} + \frac{3}{10} = \frac{4}{30} + \frac{9}{30} = \frac{13}{30}$$

$$14) \frac{7}{36} + \frac{4}{9} = \frac{7}{36} + \frac{16}{36} = \frac{23}{36}$$

$$15) \frac{6}{28} + \frac{3}{4} = \frac{6}{28} + \frac{21}{28} = \frac{27}{28}$$

$$16) \frac{12}{35} + \frac{4}{7} = \frac{12}{35} + \frac{20}{35} = \frac{32}{35}$$

$$17) \frac{2}{5} + \frac{8}{25} = \frac{10}{25} + \frac{8}{25} = \frac{18}{25}$$

$$18) \frac{7}{24} + \frac{2}{3} = \frac{7}{24} + \frac{16}{24} = \frac{23}{24}$$

Can you spot the largest fraction sum?

The largest sum is $\frac{27}{28}$



Fraction Questions And Answers For Grade 6

Lauren Gardner



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the center and is bonded to two hydrogen atoms the oxygen atom has two lone pairs of electrons this structure follows the
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web lewis structures represent the valence electrons present in the molecule as lone pairs unshared electrons drawn as dots
or bonds shared electrons drawn as lines procedure for drawing lewis structures add up the total number of valence
electrons that each atom contributes to the molecule polyatomic ion

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web the pcl₃ lewis structure has the typical case of phosphorus p in the center with 3 bonds to 3 other atoms phosphorus is
from the same column as nitrogen in the periodic table meaning that p and n generally have the same bonding structure note
the lone pair dots without bonds on top of p just like for n in the previous example for nh₃ chlorine cl

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web step 1 figure out how many electrons the molecule must have based on the number of valence electrons in each atom
when drawing the structure of an ion be sure to add subtract electrons to account for the charge step 2 connect the atoms to
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