

## 15.2 Hydroxides in Supercritical Water

### Key Concepts

Key characteristics of Supercritical Water: Dissolved salts undergo no structural changes upon heating.

Water expands, becomes aprotic, and hydroxide concentration increases with increasing temperature. Dissolved salts undergo no structural changes.

### TABLE 15.2.10 Hydroxide Chemistry

Water acts as a hydroxide source in just one sense. It can either provide a weak acid or base pair and participate itself. One direction is the other for you.

Type	Definition	Chemical Reaction
Hydroxide Ion	A common and powerful base when dissolved in water.	
Water	The process of adding or removing hydrogen atoms to form hydroxide.	
Hydroxide	Water acts as a hydroxide source in just one sense. It can either provide a weak acid or base pair and participate itself. One direction is the other for you.	
Hydroxide	The hydroxide ion is a common and powerful base when dissolved in water.	
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Hydroxide	The hydroxide ion is a common and powerful base when dissolved in water.	

# 3 2 Systems In Environmental Science

**R Bogdan**



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