

5. When neutral atoms **gain** one or more electrons, _____ are formed.
- hadrons
 - cations
 - fermions
 - deuterions
 - anions**
6. Which two of the following represent a **pair of isotopes**?
- | | | | | |
|----------------------|----------------------|----------------------|----------------------|----------------------|
| $^{31}_{12}\text{P}$ | $^{32}_{16}\text{P}$ | $^{32}_{15}\text{P}$ | $^{32}_{16}\text{S}$ | $^{33}_{16}\text{S}$ |
| I. | II. | III. | IV. | V. |
- I and IV
 - III and IV
 - I and III**
 - I and II
 - II and V
7. Which of the following consists of a pair of **allotropes**?
- air and oxygen
 - phosgene, $\text{O}=\text{C}-(\text{Cl})_2$ and phosphine, PH_3
 - red phosphorus and white phosphorus**
 - carbon tetrachloride and carbon tetrabromide
 - yellow sulfur and yellow phosphorus
8. An element that has 24 protons, 24 electrons, and 28 neutrons has a molar mass of _____.
- 76 amu.
 - 56 amu.
 - 52 amu.**
 - 48 amu.
 - 84 amu.
9. Three elements that are likely to have **similar** chemical and physical properties are:
- Ca, Co, Cr
 - H, He, Ne
 - Pb, Bi, Po
 - B, C, Si
 - Na, K, Li**
10. Which formula and name combination is **incorrect**?
- SF_6 and sulfur hexafluoride
 - N_2O_5 and dinitrogen tetraoxide**
 - P_4O_{10} and tetraphosphorus decaoxide
 - AsBr_5 and arsenic pentabromide
 - SeF_4 and selenium tetrafluoride

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1. Understanding the eBook Study Questions Chem 101 Lacc Weebly
 - The Rise of Digital Reading Study Questions Chem 101 Lacc Weebly
 - Advantages of eBooks Over Traditional Books
2. Identifying Study Questions Chem 101 Lacc Weebly
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Study Questions Chem 101 Lacc Weebly
 - User-Friendly Interface
4. Exploring eBook Recommendations from Study Questions Chem 101 Lacc Weebly
 - Personalized Recommendations
 - Study Questions Chem 101 Lacc Weebly User Reviews and Ratings
 - Study Questions Chem 101 Lacc Weebly and Bestseller Lists

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 - Study Questions Chem 101 Lacc Weebly Enhanced eBook Features
7. Enhancing Your Reading Experience
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 - Highlighting and Note-Taking Study Questions Chem 101 Lacc Weebly
 - Interactive Elements Study Questions Chem 101 Lacc Weebly
8. Staying Engaged with Study Questions Chem 101 Lacc Weebly
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Study Questions Chem 101 Lacc Weebly
9. Balancing eBooks and Physical Books Study Questions Chem 101 Lacc Weebly
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Study Questions Chem 101 Lacc Weebly
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Study Questions Chem 101 Lacc Weebly
 - Setting Reading Goals Study Questions Chem 101 Lacc Weebly
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Study Questions Chem 101 Lacc Weebly
 - Fact-Checking eBook Content of Study Questions Chem 101 Lacc Weebly
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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