

Introduction

This exam is a closed book, closed notes, no calculator, and no internet access. It is designed to test your understanding of the basic principles of chemistry, including the structure of matter, the periodic table, and the laws of thermodynamics. You are allowed to use a calculator for calculations, but you must show your work for all problems. This exam is part of the basic chemistry course, which is a prerequisite for all other chemistry courses.

Instructions: Read the questions carefully and answer them to the best of your ability. Show all your work for all problems. The exam is worth a total of 100 marks. Good luck!

Part 1: Multiple Choice Questions



1. Which of the following is not a chemical property?
 - a. Flammability
 - b. Color
 - c. Reactivity
 - d. Solubility
2. The chemical formula for water is H_2O . How many oxygen atoms are present in one molecule of water?
 - a. 1
 - b. 2
 - c. 3
 - d. 4



3. Which of the following is not a chemical property?
 - a. Flammability
 - b. Color
 - c. Reactivity
 - d. Solubility
4. The chemical formula for water is H_2O . How many oxygen atoms are present in one molecule of water?
 - a. 1
 - b. 2
 - c. 3
 - d. 4

Question	Answer	Score	Marked	Correct Answer	Incorrect Answer
1	Color	100%	Correct	Color	Flammability
2	2	100%	Correct	2	1
3	Color	100%	Correct	Color	Flammability
4	2	100%	Correct	2	1

5. The chemical formula for water is H_2O . How many oxygen atoms are present in one molecule of water?

- a. 1
- b. 2
- c. 3
- d. 4

6. The chemical formula for water is H_2O . How many oxygen atoms are present in one molecule of water?

- a. 1
- b. 2
- c. 3
- d. 4

Part 2: Short Answer Questions



1. Write the chemical formula for water.
2. Write the chemical formula for carbon dioxide.

3. Complete the following table with the appropriate values for the elements listed.

Element	Atomic Number	Mass Number	Protons	Electrons
H	1	1	1	1
O	8	16	8	8
C	6	12	6	6
Na	11	23	11	11

4. Write the chemical formula for the compound formed by the following elements: _____

Stoichiometry Phet Lab Answers

Dietmar Kennepohl



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