

Standard Thermodynamic Values at 25.0°C (298.15 K)

Formula	ΔH_f° (kJ/mol)	ΔG_f° (kJ/mol)	S_f° (J/K mol)
Ag ₂	0	0	42.55
Ag ₂ (aq)	284.55	245.65	172.997
Ag(NH ₃) ₂ ⁺ (aq)	-111.29	-17.12	245.2
Ag ₂ (S ₂ O ₃) ₂ ⁻ (aq)	-1265.7	-1033.65	98.92
Ag ⁺ (aq)	105.579	77.107	72.68
Ag ⁺ (aq)	1021.73	—	—
Ag ₂ CO ₃ (s)	-505.8	-436.8	167.4
Ag ₂ CrO ₄ (s)	-731.74	-641.76	217.6
Ag ₂ O ₂ (s)	-31.05	-11.2	121.3
Ag ₂ S(s, argenite)	-32.59	-40.67	144.01
Ag ₂ SO ₄ (s)	-715.68	-618.41	200.4
Ag ₃ PO ₄ (s)	—	-679	—
AgBr	-100.37	-96.9	107.1
AgCl(s, cerargyrite)	-137.069	-109.789	96.2
AgCl ₂ ⁻ (aq)	-245.2	-215.4	231.4
AgCN	146	156.9	107.19
AgCNS	87.9	101.39	131
Ag ₂ ⁺	-61.83	-66.19	115.5
AgNO ₃ (s)	-124.39	-33.47	140.92
Al	0	0	28.33
Al ₂ (aq)	326.4	265.7	164.54
Al(OH) ₃ (s)	-1276	—	—
Al ₂ O ₃ (s, alumina, alpha)	-1675.7	-1582.3	50.92
Al ³⁺ (aq)	-531	-485	321.7
Al ³⁺ (aq)	5483.17	—	—
AlCl ₃ (s)	-704.2	-628.8	110.67
AlCl ₃ (aq)	-583.2	—	—
Ar	0	0	154.843
As(s, gray, metallic)	0	0	35.1
B	0	0	5.86
Ba	0	0	62.6
Ba ²⁺ (aq)	-537.64	-560.77	9.6
BaC ₂ O ₄ (s)	-1368.6	—	—
BaCO ₃ (s, witherite)	-1218.3	-1137.6	112.1
BaCrO ₄ (s)	-1446	-1345.22	158.6
BaF ₂ (s)	-1207.1	-1156.6	96.36
BaSO ₄ (s)	-1473.2	-1362.2	132.2
Be	0	0	9.5
BeO	-609.6	-593.3	14.14
BF ₃ (aq)	-1137	-1120.35	254.01
Bi	0	0	96.74
Bi ₂ S ₃ (s)	-143.1	-140.6	200.4
Bi ³⁺ (aq)	—	82.8	—
Br	-121.55	-103.96	82.4
Br	111.88	62.429	174.91
Br	-219.07	—	—
Br ₂ (aq)	30.907	3.11	245.463
Br ₂ (l)	0	0	152.231
Br ₂ (g)	-130.42	-107.05	215.5
BrO ₂ (aq)	-67.07	18.6	161.71
C(s, diamond)	1.895	2.9	2.377
C(s, graphite)	0	0	5.74
C ₂	716.682	671.297	158.096
(CH ₃) ₂ O	-184.05	-112.59	266.38
C ₂ H ₂ (g, ethyne)	226.73	209.2	200.94

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