

Name: \_\_\_\_\_ Period: \_\_\_\_\_

## Properties of Water Worksheet

Word Bank:

covalent      solvent      deposition      dissolve      cohesion      surface      cooler      universal  
tension      polar      viscosity      condensation      negatively      adhesion      positively      sublimation

1. The hydrogen and oxygen atoms are held together by covalent bonds.
2. The electrons are not shared equally creating a polar molecule.
3. The polarity of water allows it to dissolve most substances. Because of this it is referred to as the universal solvent.
4. Water molecules stick to other water molecules. This property is called cohesion.
5. Hydrogen bonds form between adjacent water molecules because the positively charged hydrogen end of one water molecule attracts the negatively charged oxygen end of another water molecule.
6. Water molecules stick to other materials due to its polar nature. This property is called adhesion.
7. surface tension creates the skin-like surface formed due to the polar nature of water.
8. viscosity is the tendency for fluids to resist flow.  
cooler water is more viscous than warmer water, allowing floating organisms to use less energy to keep from sinking.
9. condensation is when water changes from a gas to a liquid.
10. sublimation is when water changes from a solid directly to a gas.
11. deposition is when water changes from a gas directly to a solid.
12. Why does ice float?  
water is less dense in solid form because hydrogen bonding between water molecules pushed molecules further apart into a set crystalline pattern
13. List the properties that ocean water has due to the fact that it contains salt.  
higher density  
conductivity  
decreased surface tension

# Properties Of Water Lab Answers

**Ralph M. Feather, Susan Leach  
Snyder, Dale T. Hesser**

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