

Assessment

Chapter Test A

Chapter: Acids and Bases

In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

- _____ 1. Which of the properties listed below is *not* characteristic of an acid?
- a sour taste
 - a slippery feel
 - the ability to conduct an electric current
 - reactivity with metals
- _____ 2. Which is an example of a weak acid?
- HCl
 - H_2SO_4
 - H_3PO_4
 - HNO_3
- _____ 3. Acetic acid is found in
- vinegar.
 - the stomach.
 - antacids.
 - oranges.
- _____ 4. A characteristic of an Arrhenius base is that it
- is an electrolyte.
 - tastes bitter.
 - dissociates to form OH^- ions.
 - All of the above
- _____ 5. Which of the following is a Brønsted-Lowry base?
- HCl
 - HCO_3^-
 - H_3O^+
 - H_3PO_4
- _____ 6. In the reaction represented by the equation
- $$\text{H}_2\text{C}_2\text{O}_4(\text{aq}) + \text{CH}_3\text{NH}_2(\text{aq}) \rightleftharpoons \text{HC}_2\text{O}_4^-(\text{aq}) + \text{CH}_3\text{NH}_3^+(\text{aq}),$$
- which of these is a conjugate acid-base pair?
- $\text{H}_2\text{C}_2\text{O}_4(\text{aq})$ and $\text{CH}_3\text{NH}_2(\text{aq})$
 - $\text{H}_2\text{C}_2\text{O}_4(\text{aq})$ and $\text{CH}_3\text{NH}_3^+(\text{aq})$
 - $\text{CH}_3\text{NH}_2(\text{aq})$ and $\text{CH}_3\text{NH}_3^+(\text{aq})$
 - $\text{HC}_2\text{O}_4^-(\text{aq})$ and $\text{CH}_3\text{NH}_3^+(\text{aq})$

Modern Chemistry Chapter 5 Test

Wilhelm Ostwald, Harry Wheeler Morse

Modern Chemistry Chapter 5 Test:

Laboratory Experiments to Accompany "Modern Chemistry," Fredus Nelson Peters,1902 **Lab Experiments for Modern Chemistry** Tzimopoulo,1990 **Lab Experiments Modern Chemistry** Harold Clark Metcalfe,1982 **Pearson Edexcel A Level Chemistry (Year 1 and Year 2)** Andrew Hunt,Graham Curtis,Graham Hill,2019-07-15 Develop and assess your students knowledge and skills throughout A level with worked examples practical assessment guidance and differentiated end of topic questions in this updated all in one textbook for Years 1 and 2 Combining everything your students need to know for the Pearson Edexcel A level Chemistry specification this revised textbook will Identify the level of your students understanding with diagnostic questions and a summary of prior knowledge at the start of the Student Book Provide support for all 16 required practicals with various activities and questions along with a Practical chapter covering procedural understanding and key ideas related to measurement Improve mathematical skills with plenty of worked examples including notes on methods to help explain the strategies for solving each type of problem Offer plenty of practice with Test yourself questions to help students assess their understanding and measure progress Encourage further reading and study with short passages of extension material Develop understanding with free online access to Test yourself answers and an extended glossary **Instructor's Manual to Accompany Chemistry in the Modern World, Concepts and Applications** Frank L. Wiseman,1985 **Handbook of Isolation and Characterization of Impurities in Pharmaceuticals** Satinder Ahuja,Karen Mills Alsante,2003-07-18 The United States Food and Drug Administration FDA and other regulatory bodies around the world require that impurities in drug substance and drug product levels recommended by the International Conference on Harmonisation ICH be isolated and characterized Identifying process related impurities and degradation products also helps us to understand the production of impurities and assists in defining degradation mechanisms When this process is performed at an early stage there is ample time to address various aspects of drug development to prevent or control the production of impurities and degradation products well before the regulatory filing and thus assure production of a high quality drug product This book therefore has been designed to meet the need for a reference text on the complex process of isolation and characterization of process related synthesis and formulation impurities and degradation products to meet critical regulatory requirements It s objective is to provide guidance on isolating and characterizing impurities of pharmaceuticals such as drug candidates drug substances and drug products The book outlines impurity identification processes and will be a key resource document for impurity analysis isolation synthesis and characterization Provides valuable information on isolation and characterization of impurities Gives a regulatory perspective on the subject Describes various considerations involved in meeting regulatory requirements Discusses various sources of impurities and degradation products *Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science* ,2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the

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