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web write the oxidation and reduction reactions for each redox reaction the first one is done for you $23 \text{ Fe} + \text{Co} \rightarrow \text{Co} + \text{Fe}$
 $\text{oxidation } \text{Co} \rightarrow \text{Co}^{2+} + 2e^{-}$ $\text{reduction } \text{Fe}^{2+} + 2e^{-} \rightarrow \text{Fe}$ $24 \text{ 3 Ag} + \text{Ni} \rightarrow \text{Ni} + 3 \text{ Ag}$ $\text{oxidation } \text{Cu} \rightarrow \text{Cu}^{2+} + 2e^{-}$ $\text{reduction } \text{Pb}^{2+} + 2e^{-} \rightarrow \text{Pb}$ 26 O_2

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 $5 \text{ CO} + \text{I}_2 \rightarrow 5 \text{ CO}_2 + \text{I}_2$

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