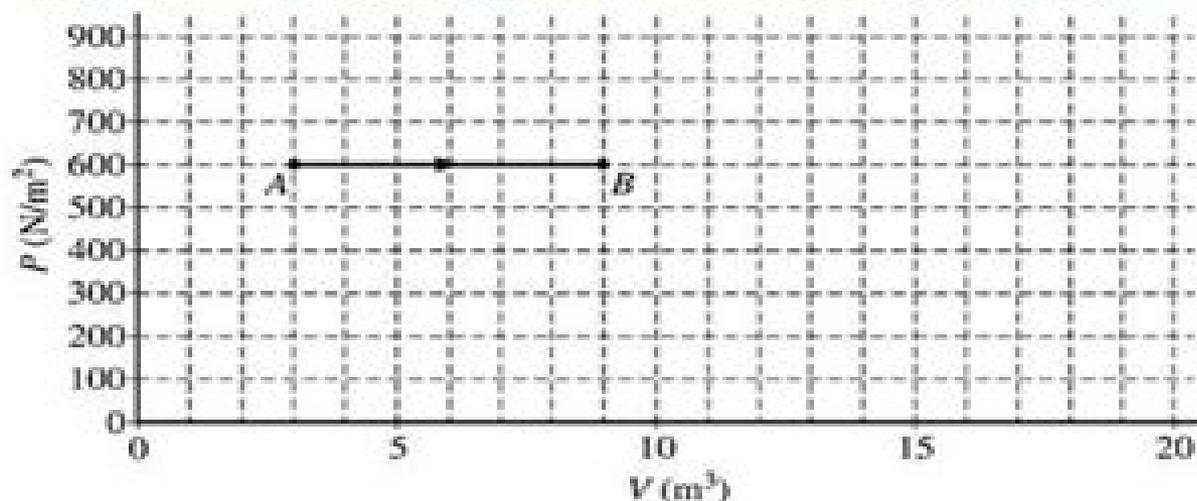


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5. (10 points)

The diagram above of pressure P versus volume V shows the expansion of 2.0 moles of a monatomic ideal gas from state A to state B . As shown in the diagram, $P_A = P_B = 600 \text{ N/m}^2$, $V_A = 3.0 \text{ m}^3$, and $V_B = 9.0 \text{ m}^3$.

(a)

- i. Calculate the work done by the gas as it expands.
- ii. Calculate the change in internal energy of the gas as it expands.
- iii. Calculate the heat added to or removed from the gas during this expansion.

(b) The pressure is then reduced to 200 N/m^2 without changing the volume as the gas is taken from state B to state C . Label state C on the diagram and draw a line or curve to represent the process from state B to state C .

(c) The gas is then compressed isothermally back to state A .

- i. Draw a line or curve on the diagram to represent this process.
- ii. Is heat added to or removed from the gas during this isothermal compression?

_____ added to _____ removed from

Justify your answer.

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Physical Review, 1999-07 Publishes papers that report results of research in statistical physics plasmas fluids and related interdisciplinary topics There are sections on 1 methods of statistical physics 2 classical fluids 3 liquid crystals 4 diffusion limited aggregation and dendritic growth 5 biological physics 6 plasma physics 7 physics of beams 8 classical physics including nonlinear media and 9 computational physics *Science* John Michels (Journalist), 1891 Vols for 1911 13 contain the Proceedings of the Helminthological Society of Washington ISSN 0018 0120 1st 15th meeting *Canadian Journal of Physics*, 2010 *Heating & Air Conditioning Contractor*, 1929 *Parliamentary Debates (Hansard)*. Great Britain. Parliament. House of Commons, 2012 **Journal of the Physical Society of Japan**, 2016 *Government Reports Announcements & Index*, 1995 **Yearbook of International Organizations 2014-2015 (Volume 4)** Union of

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