
AP CHEMISTRY FINAL EXAM REVIEW 2

1. Is energy absorbed or released in an endothermic reaction?: Absorbed.
2. What is the sign of delta H for an exothermic reaction?: Exothermic reactions release energy. The sign of delta H is negative.
3. When you touch a flask and it feels hot, is this an endothermic or exothermic process?: Exothermic; heat is being given off.
4. Define specific heat capacity.: The amount of heat required to raise 1g of a substance to 1 degrees Celsius.
5. $q=mc(\Delta T)$; What does "q" stand for?: Heat (kJ)
6. $q=mc(\Delta T)$; What does "c" stand for?: Specific heat capacity (J / g degrees C)
7. What is a hydrogen ion donor called?: An acid.
8. What is a hydrogen ion acceptor called?: A base.
9. Feels slippery; acid or base?: Base.
10. Tastes sour; acid or base?: Acid.
11. Tastes bitter; acid or base?: Base.
12. $\text{pH} < 7$; acid, base, or neutral?: Acid
13. $\text{pH} > 7$; acid, base, or neutral?: Base
14. Define neutralization.: When an acid reacts with a base to form water and a salt, $\text{pH} = 7$.
15. Soap; acid or base?: Base (tastes bitter)
16. Ammonia, acid or base?: Base (drain cleaners are bases).
17. Lemonade; acid or base?: Acid (tastes sour)
18. How many valence electrons are in an atom of chlorine?: 7
19. How many valence electrons does aluminum have?: 3
20. Define Reaction Rate: The decrease in concentration of reactants or the increase in concentration of products with time.
21. How does increasing temperature affect the rate of reaction?: Increasing temperature causes the kinetic energy to increase. Thus, molecules are moving faster and colliding more frequently. This causes the rate of reaction to increase.
22. If I decrease the concentration of reactant, what happens to the rate of reaction?: Decreasing concentration leads to less reactants available for collision. Therefore, reaction rate decreases.
23. How does surface area influence reaction rate?: By increasing the surface area of reactant molecules, we are increasing the space or point of contact that these molecules have available

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Neil D. Jespersen, Pamela Kerrigan

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Table of Contents Pre Ap Chemistry Spring Final Review Tenpayore

1. Understanding the eBook Pre Ap Chemistry Spring Final Review Tenpayore
 - The Rise of Digital Reading Pre Ap Chemistry Spring Final Review Tenpayore
 - Advantages of eBooks Over Traditional Books
2. Identifying Pre Ap Chemistry Spring Final Review Tenpayore
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Pre Ap Chemistry Spring Final Review Tenpayore
 - User-Friendly Interface
4. Exploring eBook Recommendations from Pre Ap Chemistry Spring Final Review Tenpayore

- Personalized Recommendations
 - Pre Ap Chemistry Spring Final Review Tenpayore User Reviews and Ratings
 - Pre Ap Chemistry Spring Final Review Tenpayore and Bestseller Lists
5. Accessing Pre Ap Chemistry Spring Final Review Tenpayore Free and Paid eBooks
 - Pre Ap Chemistry Spring Final Review Tenpayore Public Domain eBooks
 - Pre Ap Chemistry Spring Final Review Tenpayore eBook Subscription Services
 - Pre Ap Chemistry Spring Final Review Tenpayore Budget-Friendly Options
 6. Navigating Pre Ap Chemistry Spring Final Review Tenpayore eBook Formats
 - ePub, PDF, MOBI, and More
 - Pre Ap Chemistry Spring Final Review Tenpayore Compatibility with Devices
 - Pre Ap Chemistry Spring Final Review Tenpayore Enhanced eBook Features
 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Pre Ap Chemistry Spring Final Review Tenpayore
 - Highlighting and Note-Taking Pre Ap Chemistry Spring Final Review Tenpayore
 - Interactive Elements Pre Ap Chemistry Spring Final Review Tenpayore
 8. Staying Engaged with Pre Ap Chemistry Spring Final Review Tenpayore
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Pre Ap Chemistry Spring Final Review Tenpayore
 9. Balancing eBooks and Physical Books Pre Ap Chemistry Spring Final Review Tenpayore
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Pre Ap Chemistry Spring Final Review Tenpayore
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Pre Ap Chemistry Spring Final Review Tenpayore
 - Setting Reading Goals Pre Ap Chemistry Spring Final Review Tenpayore
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Pre Ap Chemistry Spring Final Review Tenpayore

- Fact-Checking eBook Content of Pre Ap Chemistry Spring Final Review Tenpayore
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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what is meant by radiation pattern

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directivity radiation efficiency beamwidth and bandwidth of an antenna 54 what are the radiation patterns for horn antenna
parabolic antenna 55 state the formula to find directivity for an antenna 56

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