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Consider the reaction



Estimate  $\Delta G^\circ$  for this reaction at each temperature. (Assume that  $\Delta H^\circ$  and  $\Delta S^\circ$  do not change too much within the given temperature range.)

Substance	$\Delta H_f^\circ$ (kJ/mol)	$S^\circ$ (J/(mol · K))
NO(g)	91.3	210.8
O <sub>2</sub> (g)	0	205.2
NO <sub>2</sub>	33.2	240.1

▼ Part A

298 K

Express your answer in kilojoules to three significant figures.

□
√
Δ
Σ
↶
↷
⊙
⊞
?

$\Delta G^\circ =$   kJ

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▼ Part B

# Pages 871 876 Answers

**Rachel Sandford**



**Pages 871 876 Answers:**

## **Pages 871 876 Answers** Book Review: Unveiling the Magic of Language

In an electronic era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Pages 871 876 Answers**," compiled by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound affect our existence. Throughout this critique, we will delve to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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