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# Menu Nutrition

## Breakfast

### Biscuit



290 Calories  
16g Fat  
32g Carbs  
5g Protein

### Biscuit With Bacon



355 Calories  
22g Fat  
32g Carbs  
8g Protein

### Biscuit With Sausage



520 Calories  
38g Fat  
32g Carbs  
16g Protein

### Egg & Cheese Biscuit



425 Calories  
26g Fat  
32g Carbs  
14g Protein

### Biscuits & Gravy



470 Calories  
29g Fat  
46g Carbs  
7g Protein

### Egg Sandwich



310 Calories  
12g Fat  
34g Carbs  
15g Protein

### Biscuit Sandwich with Bacon



500 Calories  
32g Fat  
32g Carbs  
18g Protein

### Biscuit Sandwich with Sausage



665 Calories  
48g Fat  
32g Carbs  
25g Protein

### Honey Butter Chicken Biscuit



560 Calories  
35g Fat  
50g Carbs  
13g Protein

### Taquito With Cheese & Potato



450 Calories  
25g Fat  
38g Carbs  
17g Protein

### Taquito With Cheese & Bacon



405 Calories  
22g Fat  
28g Carbs  
20g Protein

### Taquito With Cheese & Sausage



440 Calories  
26g Fat  
28g Carbs  
19g Protein

### Pancakes



630 Calories  
17g Fat  
108g Carbs  
9g Protein

### Breakfast Bowl with Bacon



775 Calories  
50g Fat  
53g Carbs  
27g Protein

### Breakfast Bowl with Sausage



940 Calories  
66g Fat  
53g Carbs  
34g Protein

### Breakfast on a Bun (Sausage)



525 Calories  
32g Fat  
34g Carbs  
26g Protein

### Breakfast Burger



670 Calories  
38g Fat  
51g Carbs  
28g Protein

### Hash Brown Sticks



190 Calories  
11g Fat  
21g Carbs  
2g Protein

# Protein Breakfast Prices

**University of Nebraska. Extension  
Service**

A red circular graphic with a gradient, appearing as a partial circle on the right side of the slide, overlapping the light blue bar.

## **Protein Breakfast Prices:**

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**Bulletin** ,1914      *Report of Committee on social betterment* United States. President's Homes Commission,1908

Report of the Connecticut Agricultural Experiment Station, New Haven. 1913-1920 ,1916      **Cincinnati Price-current** ,1906      **American Academy of Applied Nutrition** ,1956      Annual Report Maine. Board of Agriculture,1903      **Nature Guiding** William Gould Vinal,1926      *The Modern Cultivation of Corn* Zeferino Domínguez,1914      **Extension Bulletin** University of Nebraska. Extension Service,1914      *A Study of the day nurseries of Philadelphia* ,1916      History of Soy Flour, Flakes and Grits (510 CE to 2019) William Shurtleff; Akiko Aoyagi,2019-02-17 The world s most comprehensive well documented and well illustrated book on this subject With extensive subject and geographic index 245 photographs and illustrations mostly color Free of charge in digital format on Google Books

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section learning objectives by the end of this section you will be able to do the following describe the structure and forces present within the nucleus explain the three types of radiation write nuclear equations associated with the various types of radioactive decay

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quantum physics yet even though it has only 10 15 of the volume of the atom the nucleus contains more than 99 9 of the mass of the atom the nucleus is very small but very massive that was not predicted try to imagine the surprise and disbelief of scientists in 1911 when ernest rutherford discovered this incredible fact

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over 3 000 different kinds of atomic nuclei have been discovered in elements that span the periodic table yet each one has unique properties some may exist forever others transform within a fraction of a second some can power satellites as they travel to the outer reaches of the solar system others have special properties that allow them to treat different forms of cancer

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radioactivity some nuclides have an atomic nucleus that decays spontaneously without any external influence known as radionuclides they transform themselves into other nuclei ionising radiation is emitted during this process this is known as radioactivity

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it some future development of atomic theory perhaps a modification of the thomson model would most likely yield a causal explanation of radioactivity in terms of the internal structure of the atom

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atomic physics deals with the atom as a system consisting of a nucleus and electrons nuclear physics deals with the nucleus as a system consisting of nucleons protons and neutrons the main difference is in the scale while the term atomic deals with 1Å 10<sup>-10</sup> m where Å is an ångström according to anders jonas Ångström the term

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