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web write the oxidation and reduction reactions for each redox reaction the first one is done for you 23 $\text{Fe} + \text{CO}_2 \rightarrow \text{FeO} + \text{C}$
oxidation $\text{C} + \text{CO}_2 \rightarrow 2\text{CO}$ reduction $\text{Fe} + 2\text{Fe} \rightarrow 3\text{Fe}$ 24 $3\text{Ag} + \text{Ni} \rightarrow 3\text{Ag} + \text{Ni}$ oxidation reduction 25 $\text{Cu} + \text{Pb} \rightarrow \text{Cu} + \text{Pb}$ oxidation reduction 26 O_2

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