

Chemical 101

Intro. to Chemical Reactions

Name: _____
 Date: _____
 Class: _____

Electrolysis Reaction

1. Write the chemical formula for the following compounds:

a) hydrogen gas: H_2 b) sodium phosphate: Na_3PO_4
 c) magnesium chloride: $MgCl_2$ d) lithium nitrate: $LiNO_3$

2. Write the correct formula for the following compounds:

a) $Ca(NO_3)_2$: calcium nitrate b) $PbCl_2$: lead(II) chloride
 c) SnH_4 : stannane d) SnF_4 : stannic fluoride

Chemical Equations: What they look like...

Here is what a chemical reaction looks like when it is written:



There are examples that you are going to describe what happens during:

Chemical equation reacts with reactant and yields product and physical state

There are also the other characteristics: stoichiometric coefficient and the physical state. Here are a few more!

Critical Thinking Questions

1. Write the following reaction, with the correct state symbols.



Correctly identify each of the correct states and write the correct chemical equation.

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