

# Chemistry

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(Chapter 10) (Haloalkanes and Haloarenes) (Intext Questions)  
(Class 12)

## Question 10.8:

In the following pairs of halogen compounds, which compound undergoes faster  $S_N1$  reaction?



## Answer 10.8:



$S_N1$  reaction proceeds via the formation of carbocation. The alkyl halide (I) is  $3^\circ$  while (II) is  $2^\circ$ . Therefore, (I) forms  $3^\circ$  carbocation while (II) forms  $2^\circ$  carbocation. Greater the stability of the carbocation, faster is the rate of  $S_N1$  reaction. Since  $3^\circ$  carbocation is more stable than  $2^\circ$  carbocation, (I), i.e. 2-chloro-2-methylpropane, undergoes faster  $S_N1$  reaction than (II) i.e., 3-chloropentane.

## (ii)



The alkyl halide (I) is  $2^\circ$  while (II) is  $1^\circ$ .  $2^\circ$  carbocation is more stable than  $1^\circ$  carbocation. Therefore, (I), 2-chloroheptane, undergoes faster  $S_N1$  reaction than (II), 1-chloroheptane.

## Question 10.9:

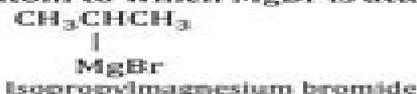
Identify A, B, C, D, E, R and R<sup>1</sup> in the following:



## Answer 10.9:



Since D of D<sub>2</sub>O gets attached to the carbon atom to which MgBr is attached, C is



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# Chemistry Chapter 10

**Lauren Gardner**



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