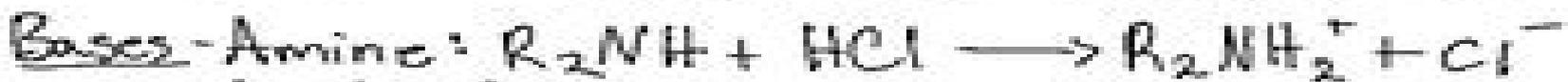


Lab Day 7 Purification by Acid-Base Extractions Prelab Questions

1) List the most common organic functional groups that function as acids or bases. Write an equation of each with HCl or NaOH to show their behavior as acids or bases. (You will also enter these functional groups in the table in Question 2.)



2) Fill in the following table showing the general solubility behavior of the organic acids and bases you listed above. Assume that they are not small enough to have significant water solubility.

Type of Organic Compound	Functional Group	Soluble in Water?	Soluble in 1 M NaOH?	Soluble in 1 M HCl?	Soluble in 1 M H ₂ O?
Neutral	Many				
Acidic	Carboxylic acids	Yes	Yes	Yes	No
Weakly Acidic	Phenols	Yes	Yes	Yes	Yes
Basic	Amines	No	Yes	Yes	Yes

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CL Gary



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Table of Contents Chem 121 Lab Manual Answers

1. Understanding the eBook Chem 121 Lab Manual Answers
 - The Rise of Digital Reading Chem 121 Lab Manual Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Chem 121 Lab Manual Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chem 121 Lab Manual Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chem 121 Lab Manual Answers
 - Personalized Recommendations
 - Chem 121 Lab Manual Answers User Reviews and Ratings
 - Chem 121 Lab Manual Answers and Bestseller Lists
5. Accessing Chem 121 Lab Manual Answers Free and Paid eBooks
 - Chem 121 Lab Manual Answers Public Domain eBooks
 - Chem 121 Lab Manual Answers eBook Subscription Services
 - Chem 121 Lab Manual Answers Budget-Friendly Options

6. Navigating Chem 121 Lab Manual Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Chem 121 Lab Manual Answers Compatibility with Devices
 - Chem 121 Lab Manual Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chem 121 Lab Manual Answers
 - Highlighting and Note-Taking Chem 121 Lab Manual Answers
 - Interactive Elements Chem 121 Lab Manual Answers
8. Staying Engaged with Chem 121 Lab Manual Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chem 121 Lab Manual Answers
9. Balancing eBooks and Physical Books Chem 121 Lab Manual Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chem 121 Lab Manual Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Chem 121 Lab Manual Answers
 - Setting Reading Goals Chem 121 Lab Manual Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chem 121 Lab Manual Answers
 - Fact-Checking eBook Content of Chem 121 Lab Manual Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

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