

# Introduction to Acid, Base, and Salt

- The nature of calcium phosphate present in tooth enamel is  
a. Basic                      b. Amphoteric                      c. Acidic                      d. Neutral  
Answer \_\_\_\_
- Which of the following salts has no water of crystallization?  
a. Blue vitriol              b. Washing soda              c. Baking soda              d. Gypsum  
Answer \_\_\_\_
- The pH of an HCl solution is 4. What is the molarity of the solution?  
a. 4.0 M                      b. 0.4 M                      c. 0.0001 M                      d. 0.001 M  
Answer \_\_\_\_
- Which of the following does not form an acidic salt?  
a. Phosphoric acid                      b. Carbonic acid  
c. Hydrochloric acid                      d. Sulfuric acid  
Answer \_\_\_\_
- The gas that is released when an acid reacts with a metal carbonate is  
a. Carbon dioxide      b. Hydrogen                      c. Oxygen                      d. Nitrogen  
Answer \_\_\_\_
- Two fruits that are rich in citric acid are  
a. Lemon and tamarind                      b. Lemon and orange  
c. Orange and tamarind                      d. Tomato and tamarind  
Answer \_\_\_\_
- The acid present in stinging hair of nettle leaves that causes burning pain  
a. Citric acid      b. Acetic acid      c. Tartaric acid      d. Methanoic acid  
Answer \_\_\_\_
- As the pH value of a solution decreases,  
a. Number of  $\text{OH}^-$  ions increases      b. Number of  $\text{H}^+$  ions increases  
c. Number of  $\text{H}^+$  ions decreases      d. Equal number of  $\text{H}^+$  ions and  $\text{OH}^-$  ions  
Answer \_\_\_\_

# Chapter 8 Solutions Acids Bases Worksheet

**Michael Seilmaier**



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