

UNIT 10: PERIODIC TABLE **PERIODIC TABLE** **PERIODIC TABLE** **PERIODIC TABLE** **PERIODIC TABLE**

Section 10.1 (continued)

In your notebook, read aloud the modern periodic table.

Copy the letterheads in the boxes on the left below from the periodic table to complete the table on the right.

atomic mass	_____
atomic number	_____
electron configuration	_____
chemical name	_____
chemical symbol	_____

Put each item in Column A, under the letter of the matching item in Column B.

Column A

Column B

- | | |
|---|----------------------------|
| _____ 18. A column in the periodic table | a. metals |
| _____ 19. A row in the periodic table | b. group |
| _____ 20. Group 1 elements | c. period |
| _____ 21. Elements that are shiny and conduct electricity | d. representative elements |
| _____ 22. Group 17 elements | e. transition elements |

In the space on the left, write each of the statements in terms of the statement in bold. Change the underlined word or phrase to make it true.

- _____ 23. There are two representative elements of platinum.
- _____ 24. Most elements besides boron of the elements in the periodic table are metalloids.
- _____ 25. Group 1A elements except for hydrogen are located on the left side.
- _____ 26. Group 1A elements are the alkali metal metals.
- _____ 27. Group 1A elements are highly reactive nonmetals known as halogens.
- _____ 28. Group 1A elements are (a) nonmetallic elements located on representative metals.
- _____ 29. Metalloids have properties of both metals and non-metallic metals.

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Excel Science Study Guide, Years 9-10 Will Marchment, 2004 The book contains coverage of five major topic areas in the NSW School Certificate test Energy Force and Motion Atoms Elements and Compounds Structure and Function of Living Things Earth and Space Ecosystems Resources and Technology a chapter on Investigations and Problem Solving in Science to help with practical skills revision questions and chapter tests to help you remember important information a glossary and summary in each section of the book diagrams and illustrations to help your understanding a section to help you prepare for the School Certificate test a sample School Certificate test paper with answers answers to all questions **DAT** Aftab S. Hassan, 1999 At Lippincott Williams and Wilkins our objective is to help busy students and actual practitioners prepare for the wide array of exams required in the health sciences Our authors and editors are skilled question writers and all our products provide the practice material that students require That's why all of our review series products the top preparatory guides formerly known as the BETZ Guides NMS BRS Recall and High Yield are the first choices of students at schools throughout the country At Lippincott Williams and Wilkins we want to help you master the science of review **Prentice Hall Chemistry**, 2000 **The Go-To Guide for Engineering Curricula, Grades 9-12** Cary I. Sneider, 2014-12-05 How to engineer change in your high school science classroom With the Next Generation Science Standards your students won't just be scientists they'll be engineers But you don't need to reinvent the wheel Seamlessly weave engineering and technology concepts into your high school math and science lessons with this collection of time tested engineering curricula for science classrooms Features include A handy table that leads you straight to the chapters you need In depth commentaries and illustrative examples A vivid picture of each curriculum its learning goals and how it addresses the NGSS More information on the integration of engineering and technology into high school science education **Understanding Organic Chemistry** Margot K. Schumm, 1987 **Resources in Education**, 1994 **Chemistry** William L. Masterton, 1996-08 This new edition of CHEMISTRY PRINCIPLES AND REACTIONS continues to provide students with the core material essential to understanding the principles of general chemistry Masterton and Hurley cover the basics without sacrificing the essentials appealing to several markets Appropriate for either a one or two semester course CHEMISTRY PRINCIPLES AND REACTIONS Fifth Edition is three hundred pages shorter than most general chemistry texts and lives up to its long standing reputation as THE student oriented text Though this text is shorter in length than most other General Chemistry books it is not lower in level and with the addition of the large volume of content provided by the revolutionary GENERAL CHEMISTRY INTERACTIVE 3.0 CD ROM that is included with every copy it has a depth and breadth rivaling much longer books **Merrill Chemistry** Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998 *Science Books & Films*, 1989 **Documentation Abstracts**, 1985 **Teaching Secondary School Science** Leslie W. Trowbridge, Rodger W. Bybee, 1996 This market leading book has been updated to reflect the latest in learning theory science reform and professional

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