

Chemical Analysis – What can you Remember?

1. What colour precipitate do chloride ions produce?
 - a) White
 - b) Cream
 - c) Yellow
 - d) Green
2. What colour do sodium ions go in a flame test?
 - a) Red
 - b) Yellow
 - c) Brick-red
 - d) Lilac
3. Which two solutions are added to test for the presence of halide ions?
 - a) Silver chloride and nitric acid
 - b) Barium chloride and hydrochloric acid
 - c) Silver nitrate and nitric acid
 - d) Barium chloride and sulphuric acid
4. What is the test for oxygen gas?
 - a) Lit splint goes out with a squeaky pop
 - b) Damp blue litmus paper goes red
 - c) Damp red litmus paper goes blue
 - d) Glowing splint relights
5. What colour precipitate do iodide ions form?
 - a) White
 - b) Cream
 - c) Yellow
 - d) Green
6. Which ion produces a brick-red colour in a flame test?
 - a) Sodium
 - b) Calcium
 - c) Copper
 - d) Lithium
7. What is the test for carbon dioxide gas?
 - a) Limewater turns cloudy
 - b) Damp blue litmus turns red
 - c) A glowing splint relights
 - d) A lit splint goes out with a squeaky pop
8. When sodium hydroxide is added to a solution containing iron (III) ions, what colour precipitate forms?
 - a) White
 - b) Blue
 - c) Green
 - d) Red/brown
9. What is the test for chlorine gas?
 - a) Damp blue litmus paper turns red and is then bleached white
 - b) Red litmus \square turns blue
 - c) A glowing splint relights
 - d) Limewater will turn cloudy
10. Copper (II) ions form what colour precipitate test?
 - a) Blue
 - b) Green
 - c) Red/brown
 - d) Cream
11. Which two solutions are added to test for the presence of sulphate ions?
 - a) Silver chloride and nitric acid
 - b) Barium chloride and sulphuric acid
 - c) Silver nitrate and nitric acid
 - d) Barium chloride and hydrochloric acid
12. What colour do potassium ions go in a flame test?
 - a) Lilac
 - b) Red
 - c) Yellow
 - d) Brick-red
13. What is the test for hydrogen gas?
 - a) Lit splint goes out with a squeaky pop
 - b) Glowing splint relights
 - c) Damp blue litmus paper turns red and then is bleached white
 - d) Limewater goes cloudy
14. Which ion produces a red colour in a flame test?
 - a) Sodium
 - b) Calcium
 - c) Copper
 - d) Lithium
15. What colour precipitate do bromide ions form?
 - a) White
 - b) Cream
 - c) Yellow
 - d) Green
16. What needs to be added to test for the presence of carbonate ions?
 - a) Silver nitrate
 - b) Universal indicator
 - c) Litmus paper
 - d) Dilute acid
17. If sodium hydroxide is added to a compound containing ammonium ions, what gas is produced?
 - a) Oxygen
 - b) Ammonia
 - c) Carbon dioxide
 - d) Hydrogen
18. What colour precipitate does barium sulphate form?
 - a) White
 - b) Green
 - c) Yellow
 - d) Cream
19. What is needed to carry out a flame test?
 - a) Splint
 - b) Glass rod
 - c) Spatula
 - d) Nichrome/platinum wire

Your score out of 19=
Percentage =

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A Quiz-system of Pharmacy and All the Allied Branches William Josephus Robinson, 1903

Study and Quiz Outline William Lloyd Evans, 1923

Fundamentals of Organic Chemistry T. W. Graham Solomons, 1993

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Laboratory Instruction in the Field of Inorganic Chemistry Victor Herbert Noll, 1930

The Temple University Catalogue Temple University, 1913

Chemtrek Stephen Thompson, 1990

Chemical Principles Peter Atkins, Loretta Jones, 2007-08 Written for calculus inclusive general chemistry courses Chemical Principles helps students develop chemical

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Table of Contents Chemistry Solutions Quiz

1. Understanding the eBook Chemistry Solutions Quiz
 - The Rise of Digital Reading Chemistry Solutions Quiz
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Solutions Quiz
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry Solutions Quiz
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Solutions Quiz
 - Personalized Recommendations
 - Chemistry Solutions Quiz User Reviews and Ratings
 - Chemistry Solutions Quiz and Bestseller Lists
5. Accessing Chemistry Solutions Quiz Free and Paid eBooks
 - Chemistry Solutions Quiz Public Domain eBooks

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- Chemistry Solutions Quiz Budget-Friendly Options
- 6. Navigating Chemistry Solutions Quiz eBook Formats
 - ePub, PDF, MOBI, and More
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 - Chemistry Solutions Quiz Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemistry Solutions Quiz
 - Highlighting and Note-Taking Chemistry Solutions Quiz
 - Interactive Elements Chemistry Solutions Quiz
- 8. Staying Engaged with Chemistry Solutions Quiz
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry Solutions Quiz
- 9. Balancing eBooks and Physical Books Chemistry Solutions Quiz
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry Solutions Quiz
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Chemistry Solutions Quiz
 - Setting Reading Goals Chemistry Solutions Quiz
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chemistry Solutions Quiz
 - Fact-Checking eBook Content of Chemistry Solutions Quiz
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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