

## PRE TEST

## Chapter 6 Test-General/Honors Chemistry

## Part 1: Matching

Match each description in Column B with the correct term in Column A. Write the letter of the correct description on the line.

- |       |              |                           |  |
|-------|--------------|---------------------------|--|
| C.2.2 | <u>  a  </u> | 1. cation                 | a. a positively charged ion  |
| C.2.2 | <u>  g  </u> | 2. electronegativity      | b. an element whose highest occupied <i>s</i> or <i>p</i> sublevels are partially filled                     |
| C.2.5 | <u>  j  </u> | 3. inner transition metal | c. the energy required to remove an electron from an atom  |
| C.2.6 | <u>  i  </u> | 4. alkali earth metal     | d. good conductor of heat and electric current   |
| C.2.6 | <u>  h  </u> | 5. noble gas              | e. any element whose highest occupied <i>s</i> sublevel and a nearby <i>d</i> sublevel contain electrons     |
| C.2.5 | <u>  b  </u> | 6. representative element | f. any nonmetal in group 7A  |
| C.2.2 | <u>  d  </u> | 7. metal                  | g. the tendency of an atom to attract electrons when it's in a compound (how much an atom wants an electron) |
| C.2.6 | <u>  f  </u> | 8. halogen                | h. an element in which the highest occupied <i>s</i> and <i>p</i> sublevels are completely filled            |
| C.2.5 | <u>  e  </u> | 9. transition metal       | i. an element in group 2A  |
| C.2.2 | <u>  c  </u> | 10. ionization energy     | j. an element whose highest <i>s</i> sublevel and a nearby <i>f</i> sublevel contain electrons               |

## Part 2: Multiple Choice

Choose the best answer and write its letter on the line.

- C.2.6   b   11. The alkali metals do not include
- |       |       |
|-------|-------|
| a. Li | c. Na |
| b. Ca | d. Rb |
- C.2.6   c   12. The element iodine, I, is a
- |                          |                              |
|--------------------------|------------------------------|
| a. period 5 alkali metal | c. period 5 halogen          |
| b. period 4 halogen      | d. period 5 transition metal |

# Chemistry Chapter 6 Test

**Jin-Ying Zhang**



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*Flue Gas Desulfurization and Industrial Minerals* M. Michael Miller, Deborah A. Kramer, G. Oliver Vagt, 1993 Contains 4 101 references on FGD Flue Gas Desulfurization primarily from 1982 through June 1993 Complements the Flue Gas Desulfurization and Denitrification bibliography published by the U S Dept of Energy in Jan 1985 References were located on the Energy Science and Technology Pollution Abstracts and Environmental Bibliography databases Primarily covers FGD and the use of industrial minerals in the desulfurization process or in by product utilization and disposal Emphasizes post combustion removal of sulfur dioxide through processes such as in duct injection and wet and dry scrubbing

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## **Table of Contents Chemistry Chapter 6 Test**

1. Understanding the eBook Chemistry Chapter 6 Test
  - The Rise of Digital Reading Chemistry Chapter 6 Test
  - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Chapter 6 Test
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Chemistry Chapter 6 Test
  - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Chapter 6 Test
  - Personalized Recommendations
  - Chemistry Chapter 6 Test User Reviews and Ratings
  - Chemistry Chapter 6 Test and Bestseller Lists
5. Accessing Chemistry Chapter 6 Test Free and Paid eBooks

- Chemistry Chapter 6 Test Public Domain eBooks
  - Chemistry Chapter 6 Test eBook Subscription Services
  - Chemistry Chapter 6 Test Budget-Friendly Options
6. Navigating Chemistry Chapter 6 Test eBook Formats
    - ePub, PDF, MOBI, and More
    - Chemistry Chapter 6 Test Compatibility with Devices
    - Chemistry Chapter 6 Test Enhanced eBook Features
  7. Enhancing Your Reading Experience
    - Adjustable Fonts and Text Sizes of Chemistry Chapter 6 Test
    - Highlighting and Note-Taking Chemistry Chapter 6 Test
    - Interactive Elements Chemistry Chapter 6 Test
  8. Staying Engaged with Chemistry Chapter 6 Test
    - Joining Online Reading Communities
    - Participating in Virtual Book Clubs
    - Following Authors and Publishers Chemistry Chapter 6 Test
  9. Balancing eBooks and Physical Books Chemistry Chapter 6 Test
    - Benefits of a Digital Library
    - Creating a Diverse Reading Collection Chemistry Chapter 6 Test
  10. Overcoming Reading Challenges
    - Dealing with Digital Eye Strain
    - Minimizing Distractions
    - Managing Screen Time
  11. Cultivating a Reading Routine Chemistry Chapter 6 Test
    - Setting Reading Goals Chemistry Chapter 6 Test
    - Carving Out Dedicated Reading Time
  12. Sourcing Reliable Information of Chemistry Chapter 6 Test
    - Fact-Checking eBook Content of Chemistry Chapter 6 Test
    - Distinguishing Credible Sources
  13. Promoting Lifelong Learning
    - Utilizing eBooks for Skill Development

- Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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