

## CHEM 100

## Principles of Chemistry

### Course Syllabus and Information

Description:	<p>This class is an introduction to chemistry with emphasis on a semi-empirical approach to the scientific method, the microscopic properties of matter, chemical reactions and chemical equations. Examples of chemistry from history, technology, biology and environmental science will be introduced. This is primarily a preparatory class for CHEM 101 and CHEM 102. <b>A grade of C or better (C- is unacceptable) in this class serves as a prerequisite for CHEM 101.</b> You can advance directly to CHEM 101 (without taking CHEM 100) if you score 40 out of 60 or better in the Chemistry Placement Test (CPT). Due to the ongoing Coronavirus pandemic, the in-person Chemistry Placement Test (CPT) has been suspended. Instead, An Interim Chemistry Placement Test (ICPT) offers a temporary and online solution. A passing score (40 out of 60) in the ICPT will allow students to satisfy the prerequisite for CHEM 101. See <a href="https://www.csun.edu/testing/chemistry-placement-test-cpt">https://www.csun.edu/testing/chemistry-placement-test-cpt</a> for more information. Credit cannot be earned in both CHEM 100 and CHEM 103. Although successful completion of this class and its associated laboratory satisfies the Subject Explorations, Natural Sciences General Education requirement, you <b>SHOULD NOT</b> take CHEM 100 to satisfy a GE requirement only – <b>TAKE CHEM 110 INSTEAD.</b></p>
Outcomes:	<p>Students successfully completing this class will be able to:</p> <ul style="list-style-type: none"><li>● Develop strategies for problem solving for various types of chemical calculations</li><li>● Describe and quantify the composition of chemical solutions</li><li>● Describe the features and application of the scientific method</li><li>● Perform calculations involving measurements and chemical quantities</li><li>● Understand how chemistry impacts everyday life</li><li>● Understand how energy is involved in light, temperature and chemical change</li><li>● Understand the composition of matter - atoms, molecules, ions and compounds - at a fundamental level</li><li>● Understand the organization of the periodic table</li><li>● Write chemical names, symbols, formulas and chemical reactions</li></ul>
Instructor:	<b>Dr. Kamatchi Lakshminarayanan</b>
E-mail:	kamatchi.lakshminarayanan@csun.edu
Prerequisites:	<p>There are several ways you can meet the prerequisites for this class:</p> <ul style="list-style-type: none"><li>● An ELM score <math>\geq 50</math></li><li>● ELM exemption through MATH 093 (CR or C grade or better)</li><li>● ELM exemption through MATH 091A (ALEKS assessment or MSA score of 70 or above)</li><li>● A passing grade in MATH 196S or MATH 196QR</li><li>● Co-enrollment in MATH 102, MATH 105, MATH 255A or MATH 150A</li></ul>
SI Class:	<p>There is a 1-unit, credit/no credit Supplemental Instruction (SI) class, UNIV 060C, that accompanies CHEM 100. The class offers more worked examples and individualized</p>

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## Unveiling the Magic of Words: A Overview of "**Chem 100 Principles Of Chemistry Course Syllabus And**"

In some sort of defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their power to kindle emotions, provoke contemplation, and ignite transformative change is actually awe-inspiring. Enter the realm of "**Chem 100 Principles Of Chemistry Course Syllabus And**," a mesmerizing literary masterpiece penned by a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve in to the book is central themes, examine its distinctive writing style, and assess its profound impact on the souls of its readers.

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