

# Atomic Structure & Periodic Table Worksheet

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1. The table below shows the structure of different atoms and ions. Using the periodic table, complete the table with missing values.

Element	Atomic Number	Proton Number	Neutron Number	Electron Number	Mass Number
Mg	12		12		
Mg <sup>2+</sup>	12				24
F			10		
F <sup>-1</sup>			10		19

2. The table below gives the composition of three different particles X, Y, and Z.

Particle	Proton Number	Neutron Number	Electron Number
X	15	16	15
Y	15	16	18
Z	15	17	15

State the evidence in the table for each of the following statements.

- a) Particle X is an atom.

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- b) Particle Y is a negative ion.

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- c) X, Y, and Z are all particles of the same element.

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- d) Particles X and Z are isotopes of the same elements.

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- e) Particles X and Z have zero charges.

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# Chapter Test Atom And Periodic Table

**SA Adler**



## **Chapter Test Atom And Periodic Table :**

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### **Table of Contents Chapter Test Atom And Periodic Table**

1. Understanding the eBook Chapter Test Atom And Periodic Table
  - The Rise of Digital Reading Chapter Test Atom And Periodic Table
  - Advantages of eBooks Over Traditional Books
2. Identifying Chapter Test Atom And Periodic Table
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Chapter Test Atom And Periodic Table
  - User-Friendly Interface
4. Exploring eBook Recommendations from Chapter Test Atom And Periodic Table
  - Personalized Recommendations

- Chapter Test Atom And Periodic Table User Reviews and Ratings
- Chapter Test Atom And Periodic Table and Bestseller Lists
- 5. Accessing Chapter Test Atom And Periodic Table Free and Paid eBooks
  - Chapter Test Atom And Periodic Table Public Domain eBooks
  - Chapter Test Atom And Periodic Table eBook Subscription Services
  - Chapter Test Atom And Periodic Table Budget-Friendly Options
- 6. Navigating Chapter Test Atom And Periodic Table eBook Formats
  - ePub, PDF, MOBI, and More
  - Chapter Test Atom And Periodic Table Compatibility with Devices
  - Chapter Test Atom And Periodic Table Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Chapter Test Atom And Periodic Table
  - Highlighting and Note-Taking Chapter Test Atom And Periodic Table
  - Interactive Elements Chapter Test Atom And Periodic Table
- 8. Staying Engaged with Chapter Test Atom And Periodic Table
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Chapter Test Atom And Periodic Table
- 9. Balancing eBooks and Physical Books Chapter Test Atom And Periodic Table
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Chapter Test Atom And Periodic Table
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Chapter Test Atom And Periodic Table
  - Setting Reading Goals Chapter Test Atom And Periodic Table
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chapter Test Atom And Periodic Table
  - Fact-Checking eBook Content of Chapter Test Atom And Periodic Table

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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