

CHEMISTRY REVIEW: KEY CHAPTER 9 – CHEMICAL NAMES AND FORMULAS

- In a Stock name such as iron (II) sulfide, the roman numeral tells us (2)
 - How many atoms of Fe are in one formula unit.
 - How many sulfide ions can be attached to the iron atom.
 - The charge on each Fe ion.
 - The total positive charge of the formula unit.
- The result of changing a subscript in a correctly written chemical formula is (1)
 - Change the number of ions represented in the formula.
 - Change the charges on the other ions in the compound.
 - Change the formula so that it no longer represents the compound.
 - Have no effect on the formula.
- Derive a generalization for when an acid name will end in the suffix -ic or -ous. If ending in -ic, the anion ended in -ate. If ending in -ous, the anion ended in -ite.
- Derive a generalization for when an acid name will begin with the prefix hydro-. If prefix is hydro-, the anion ended in -ite. This is almost always a monatomic ion.
- Fill in the blanks in the table below.

Compound Name	Formula
Acetic acid	$\text{CH}_3\text{CO}_2\text{H}$
Aluminum sulfate	$\text{Al}_2(\text{SO}_4)_3$
Lead II carbonate	PbCO_3
Ammonium phosphate	$(\text{NH}_4)_3\text{PO}_4$
Sulfuric acid	H_2SO_4
Nitrogen pentoxide	N_2O_5
Copper II carbonate	CuCO_3
Sodium nitrate	NaNO_3
Ammonium phosphate	$(\text{NH}_4)_3\text{PO}_4$
Mercury sulfate	HgSO_4
Hydroic acid	HNO_3
Magnesium perchlorate	$\text{Mg}(\text{ClO}_4)_2$
Iron II chloride	FeCl_2
Iron III chloride	FeCl_3
Carbon Dioxide	CO_2
Sulfur dioxide	SO_2
Sulfur pentachloride	SCl_5
Tetraphosphorus hexoxide	P_4O_6
Tetrasulfur dioxide	S_4O_2

- Name the following compound and identify its molecular ions or acid.
 - $\text{Ca}(\text{NO}_3)_2$, calcium (II) nitrate - II
 - $\text{Fe}(\text{NO}_3)_3$, Iron (III) nitrate - III
 - $\text{Fe}(\text{OH})_2$, Iron (II) hydroxide - II
 - N_2O_5 , Nitrogen pentoxide - III

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Ying-Ying Zheng



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High School Chemistry Michael A. Buben, 2018-05-18 The Teacher's Edition is designed to add direction to High School Chemistry which is an outline of notes. The books go hand in hand. Modified Second Edition as a student's workbook and guidance from the Teacher's Edition. **Basic Chemical Principles** Donada Peters, Edward I. Peters, 1988. General, Organic, and Biological Chemistry John R. Amend, Bradford P. Mundy, Melvin T. Arnold, 1990. Very Good. No Highlights or Markup. All pages are intact. **Journal of Chemical Education**, 1929. Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science, 2003-11. Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day. Relevant content, lively explorations, and a wealth of hands-on activities take students' understanding of science beyond the page and into the world around them. Now includes even more technology tools and activities to support differentiated instruction. NIOSH Pocket Guide to Chemical Hazards, 1994. The NIOSH Pocket Guide to Chemical Hazards NPG is intended as a source of general industrial hygiene information on several hundred chemicals classes for workers, employers, and occupational health professionals. The guide presents key information and data in abbreviated or tabular form for chemicals or substance groupings, e.g., cyanides, fluorides, manganese compounds that are found in the work environment. The guide's introduction. **Chemistry: The Study of Matter** Henry Dorin, 1982. **Chemical news and Journal of physical science**, 1898. **The Chemical News and Journal of Physical Science**, 1913. The Druggists' Circular and Chemical Gazette, 1894. Includes Red book price list section. Title varies slightly. Issued semiannually 1897-1906. **Chemical News and Journal of Industrial Science**, 1882. **Journal of the Society of Chemical Industry** Society of Chemical Industry (Great Britain), 1906. Includes list of members 1882-1902 and proceedings of the annual meetings and various supplements. British Chemical Abstracts, 1926. American Professional Pharmacist, 1947. **The Chemical News and Journal of Industrial Science**, 1913. The American Perfumer, 1939. English Mechanic and World of Science, 1884. **Merck's Report** Theodore Weicker, 1894. **Proceedings of the American Pharmaceutical Association**, 1930. Stirling Stirling Consolidated Boiler Company, 1906.

Unveiling the Energy of Verbal Artistry: An Psychological Sojourn through **Chapter 9 Chemical Names Formulas Answers Page 221**

In some sort of inundated with screens and the cacophony of immediate conversation, the profound power and emotional resonance of verbal beauty usually disappear into obscurity, eclipsed by the continuous assault of sound and distractions. Yet, nestled within the musical pages of **Chapter 9 Chemical Names Formulas Answers Page 221**, a interesting perform of literary splendor that pulses with natural thoughts, lies an wonderful journey waiting to be embarked upon. Composed with a virtuoso wordsmith, this mesmerizing opus guides viewers on a mental odyssey, lightly exposing the latent possible and profound influence stuck within the complex web of language. Within the heart-wrenching expanse of the evocative evaluation, we shall embark upon an introspective exploration of the book is central styles, dissect their interesting publishing style, and immerse ourselves in the indelible impact it leaves upon the depths of readers souls.

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