

Calculate the mass of 3.011×10^{23} number of oxygen atoms.

$$\text{Number of moles} = \frac{\text{Number of particles}}{\text{Avogadro number}}$$

$$\text{Number of moles} = \frac{3.011 \times 10^{23}}{6.022 \times 10^{23}}$$

$$= \frac{1}{2} = 0.5 \text{ mole}$$

Chapter 11 The Mole Answer Key

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