

## Chemistry Worksheets Class 10 on Chapter 1 Chemical Reactions and Equations- Set 1

**Q-1:** Write the balanced chemical equations for each of the following reactions.

- Liquid hydrogen peroxide decomposes to form water and oxygen gas.
- Acetylene gas burns in oxygen to form carbon dioxide and water along with evolution of heat.
- When solid mercury(II) oxide is heated, liquid mercury and oxygen gas are produced.
- Marble dissolves in a hydrochloric acid to give calcium chloride, water and carbon dioxide.
- Copper sulphate on treatment with potassium iodide precipitates cuprous iodide, liberates iodine gas and also forms potassium sulphate.

**Q-2:** Fill in the blanks

- \_\_\_\_\_ is the magnetic oxide of iron.
- In reactivity series, manganese lies \_\_\_\_\_ zinc.
- To the reaction between  $\text{AgNO}_3$  solution and  $\text{NaCl}$  solution, the names given to this type of reactions are \_\_\_\_\_ and \_\_\_\_\_.
- Proteins decompose to form \_\_\_\_\_ in our body.
- If copper turnings are added to colourless  $\text{AgNO}_3$  solution, it turns \_\_\_\_\_ after sometime.

**Q-3:** Which of the following displacement reactions will not take place?

- $\text{Zn(s)} + \text{FeSO}_4(\text{aq}) \rightarrow \text{ZnSO}_4(\text{aq}) + \text{Fe(s)}$
- $\text{Fe(s)} + \text{CuSO}_4(\text{aq}) \rightarrow \text{FeSO}_4(\text{aq}) + \text{Cu(s)}$
- $\text{Cu(s)} + 2\text{AgNO}_3(\text{aq}) \rightarrow \text{Cu(NO}_3)_2(\text{aq}) + 2\text{Ag(s)}$
- $\text{Cu(s)} + \text{FeSO}_4(\text{aq}) \rightarrow \text{CuSO}_4(\text{aq}) + \text{Fe(s)}$

**Q-4:** Which of the following statements is false?

- Oxidation results in the loss of an electropositive element and the gain of an electronegative element.
- Reduction results in the loss of an electronegative element and the gain of an electropositive element.
- A species that gains electrons is referred to as a reducing agent.
- An oxidising agent is a type of specie that absorbs hydrogen atoms.

**Q-5:** Why are certain chemicals and medications kept in coloured bottles?

**Q-6:** Sakshi desired that her home be white washed. She went to the market and bought 20 kg of quick lime, which she dissolved in 60 litres of water. When she added lime to water, she noticed that the water began to boil even though it was not heated. Give an explanation for her observation. Name the product and write the corresponding chemical equation and word equation.

# Chapter 10 Chemistry Answers

**Hussin A.Rothana**



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## **Table of Contents Chapter 10 Chemistry Answers**

1. Understanding the eBook Chapter 10 Chemistry Answers
  - The Rise of Digital Reading Chapter 10 Chemistry Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Chapter 10 Chemistry Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Chapter 10 Chemistry Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Chapter 10 Chemistry Answers
  - Personalized Recommendations
  - Chapter 10 Chemistry Answers User Reviews and Ratings
  - Chapter 10 Chemistry Answers and Bestseller Lists
5. Accessing Chapter 10 Chemistry Answers Free and Paid eBooks
  - Chapter 10 Chemistry Answers Public Domain eBooks
  - Chapter 10 Chemistry Answers eBook Subscription Services
  - Chapter 10 Chemistry Answers Budget-Friendly Options
6. Navigating Chapter 10 Chemistry Answers eBook Formats

- ePub, PDF, MOBI, and More
  - Chapter 10 Chemistry Answers Compatibility with Devices
  - Chapter 10 Chemistry Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
    - Adjustable Fonts and Text Sizes of Chapter 10 Chemistry Answers
    - Highlighting and Note-Taking Chapter 10 Chemistry Answers
    - Interactive Elements Chapter 10 Chemistry Answers
  8. Staying Engaged with Chapter 10 Chemistry Answers
    - Joining Online Reading Communities
    - Participating in Virtual Book Clubs
    - Following Authors and Publishers Chapter 10 Chemistry Answers
  9. Balancing eBooks and Physical Books Chapter 10 Chemistry Answers
    - Benefits of a Digital Library
    - Creating a Diverse Reading Collection Chapter 10 Chemistry Answers
  10. Overcoming Reading Challenges
    - Dealing with Digital Eye Strain
    - Minimizing Distractions
    - Managing Screen Time
  11. Cultivating a Reading Routine Chapter 10 Chemistry Answers
    - Setting Reading Goals Chapter 10 Chemistry Answers
    - Carving Out Dedicated Reading Time
  12. Sourcing Reliable Information of Chapter 10 Chemistry Answers
    - Fact-Checking eBook Content of Chapter 10 Chemistry Answers
    - Distinguishing Credible Sources
  13. Promoting Lifelong Learning
    - Utilizing eBooks for Skill Development
    - Exploring Educational eBooks
  14. Embracing eBook Trends
    - Integration of Multimedia Elements
    - Interactive and Gamified eBooks

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