

# The Chemistry of Cookies

Get this "bake" gone with the chemistry lab. Students investigate the role of each main ingredient & how conditions such as baking temperature and time affect the appearance and taste of the cookies. This will be a fun and educational activity for students. Includes a list of ingredients, a list of equipment, and a list of safety precautions.

**Table 1: Ingredient List** (Note: The following table lists the ingredients for the cookies, their chemical formulas, and their molar masses. The molar masses are listed in g/mol.)

Ingredient	Chemical Formula	Molar Mass (g/mol)	Ingredient	Molar Mass (g/mol)	
Flour	$C_6H_{10}O_5$	162.14	Vanilla	$C_{10}H_{14}O_2$	166.20
Baking Soda	$NaHCO_3$	84.01	Egg	$C_5H_7NO_2$	133.07
Baking Powder	$NaHCO_3$	84.01	Butter	$C_{55}H_{98}O_6$	901.30
Brown Sugar	$C_{12}H_{22}O_{11}$	342.30	Salt	$NaCl$	58.44
Butter	$C_{55}H_{98}O_6$	901.30	Chocolate	$C_{36}H_{24}O_7$	512.52
Vanilla Sugar	$C_{10}H_{14}O_2$	166.20	Egg	$C_5H_7NO_2$	133.07
Chocolate Chips	$C_{36}H_{24}O_7$	512.52			
Confectioner's Sugar	$C_{12}H_{22}O_{11}$	342.30			
Egg	$C_5H_7NO_2$	133.07			

**Table 2: Recipe Instructions** (Note: The following table lists the ingredients for the cookies, their quantities, and their molar masses. The molar masses are listed in g/mol.)

Ingredient	Quantity	Molar Mass (g/mol)
Flour	2 cups (250 g)	162.14
Baking Soda	1/2 tsp (1.25 g)	84.01
Baking Powder	1/2 tsp (1.25 g)	84.01
Brown Sugar	1 cup (200 g)	342.30
Butter	1/2 cup (113 g)	901.30
Vanilla Sugar	1/2 cup (75 g)	166.20
Chocolate Chips	1/2 cup (100 g)	512.52
Confectioner's Sugar	1/2 cup (75 g)	342.30
Egg	2 eggs (100 g)	133.07

## Ingredients

- 2 cups (250 g) flour
- 1/2 tsp (1.25 g) baking soda
- 1/2 tsp (1.25 g) baking powder
- 1 cup (200 g) brown sugar
- 1/2 cup (113 g) butter
- 1/2 cup (75 g) vanilla sugar
- 1/2 cup (100 g) chocolate chips
- 1/2 cup (75 g) confectioner's sugar
- 2 eggs (100 g)



# Chemistry Chocolate Chip Cookies Answers

**Arielle Johnson**



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