

Chapter Test B

Chapter: Chemical Equations and Reactions

PART I In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

- _____ 1. The production of a slightly soluble solid compound in a double-displacement reaction results in the formation of a
- gas.
 - precipitate.
 - combustion reaction.
 - halogen.
- _____ 2. To balance a chemical equation, it is permissible to adjust the
- coefficients.
 - subscripts.
 - formulas of the products.
 - number of products.
- _____ 3. In a chemical equation, the symbol (aq) indicates that the substance is
- water.
 - dissolved in water.
 - an acid.
 - insoluble.
- _____ 4. The tendency for a replacement reaction to occur increases as the
- interval between any two elements in the activity series decreases.
 - temperature decreases.
 - valence electrons are used up.
 - interval between any two elements in the activity series increases.
- _____ 5. The coefficients in a chemical equation
- indicate the number of moles of each substance that react.
 - show the number of grams of each substance that react.
 - are the molar masses of the substances.
 - show the valence electrons for each atom.
- _____ 6. If metal X is lower than metal Y in the activity series, then
- X will replace ions of Y in solution.
 - Y will replace ions of X in solution.
 - Y will form oxides only indirectly.
 - X will react with cold water.

Chapter Test B Chemical Reactions Answers

K Morrison



Chapter Test B Chemical Reactions Answers:

Chapter Test B Chemical Reactions Answers Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Chapter Test B Chemical Reactions Answers**," written by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we will delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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