

Name \_\_\_\_\_ Date \_\_\_\_\_ Class \_\_\_\_\_

## CHAPTER 9

## STUDY GUIDE FOR CONTENT MASTERY

## Covalent Bonding

### Section 11 The Covalent Bond

In your textbook, read about the nature of covalent bonds.

Use each of the terms below just once to complete the passage.

covalent bond	molecule	sigma bond	exothermic	$\pi$ bond
---------------	----------	------------	------------	------------

When sharing electrons occurs, the attachment between atoms that results is called a(n) **covalent bond**. When such an attachment is formed, bond dissociation energy is released, and the process is **exothermic**. When two or more atoms bond by means of electron sharing, the resulting particle is called a(n) **molecule**. If the electrons shared are contained between the two atoms, the attachment is called a(n) **sigma bond**. If the sharing involves the overlap of parallel orbitals, the attachment is called a(n)  **$\pi$  bond**.

In your textbook, read about single and multiple bonds and bond strengths.

Circle the letter of the choice that best completes the statement or answers the question.

- Is a water molecule a(n) **covalent bond**, **sigma bond**, **exothermic**, or  **$\pi$  bond**?
  - an ionic atom
  - a molecule containing three atoms
  - a molecule containing two atoms
  - a molecule containing four atoms
- How many electrons are shared in a double covalent bond?
  - none
  - two
  - four
  - eight
- Bond length is the distance between
  - two molecules of the same substance
  - the electron in two attached atoms
  - the nuclei of two attached atoms
  - the orbitals of two attached atoms
- Which of the following relationships relating to bond length is generally correct?
  - the shorter the bond, the stronger the bond
  - the shorter the bond, the weaker the bond
  - the shorter the bond, the fewer the electrons in it
  - the shorter the bond, the lower the bond dissociation energy

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Chemistry: Matter and Change • Chapter 9

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## CHAPTER 9 STUDY GUIDE FOR CONTENT MASTERY

### Section 12 Naming Molecules

In your textbook, read about how binary compounds and acids are named from their formulas.

For each statement below, write *true* or *false*.

- false** \_\_\_\_\_ 1. Binary molecular compounds are generally composed of a metal and a nonmetal.
- false** \_\_\_\_\_ 2. The second element in the formula of a binary compound is named using the suffix *-ite*.
- false** \_\_\_\_\_ 3. The prefix *mono-* indicates three atoms.
- true** \_\_\_\_\_ 4. The prefix *hepta-* indicates six atoms.
- true** \_\_\_\_\_ 5. In naming the first element in a formula, the prefix *mono-* is not used.
- true** \_\_\_\_\_ 6. For binary acids, the hydrogen part of the compound is named using the prefix *hydro-*.
- false** \_\_\_\_\_ 7. An oxyacid contains only two elements.
- false** \_\_\_\_\_ 8. If the name of the cation of an oxyacid ends in *-ite*, the acid name contains the suffix *-ate*.

In your textbook, read about naming molecular compounds and oxyacids.

For each item in Column A, write the letter of the matching item in Column B.

## Column A

- c.** \_\_\_\_\_ 9. CO
- f.** \_\_\_\_\_ 10. CO<sub>2</sub>
- g.** \_\_\_\_\_ 11. H<sub>2</sub>CO<sub>3</sub>
- e.** \_\_\_\_\_ 12. N<sub>2</sub>
- b.** \_\_\_\_\_ 13. N<sub>2</sub>O<sub>4</sub>
- d.** \_\_\_\_\_ 14. HNO<sub>3</sub>
- f.** \_\_\_\_\_ 15. HNO<sub>2</sub>
- a.** \_\_\_\_\_ 16. HBr
- h.** \_\_\_\_\_ 17. HBrO<sub>3</sub>

## Column B

- a.** hydrobromic acid
- b.** nitrogen tetroxide
- c.** carbon monoxide
- d.** nitrous acid
- e.** carbonic acid
- f.** nitric acid
- g.** carbonic acid
- h.** bromic acid
- i.** carbon dioxide

Study Guide for Content Mastery

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# Chapter 7 Study Guide For Content Mastery Answer Key

**Yijin Wang**



**Chapter 7 Study Guide For Content Mastery Answer Key:**

## Decoding **Chapter 7 Study Guide For Content Mastery Answer Key**: Revealing the Captivating Potential of Verbal Expression

In a period characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its power to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Chapter 7 Study Guide For Content Mastery Answer Key**," a mesmerizing literary creation penned with a celebrated wordsmith, readers attempt an enlightening odyssey, unraveling the intricate significance of language and its enduring effect on our lives. In this appraisal, we shall explore the book's central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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web sep 9 2018 issue 2 many time we see the performance of the fin fan cooler is not as ok and unable to maintain the proper cooling it is because of the many reason e g fouling of the tubes inside of the

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web life of the cooler this is the preferred tube for operating temperatures up to 600 f the embedded fin also affords a continued predictable heat transfer and should be used for all coolers operating above 600 f and below 750 f the wrap on footed fin tube can be used below 250 f however the bond between the fin

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web introduction fin fan cooler is a kind of heat exchanger that uses the air for cooling medium it is required a wide surface area for heat transfer it is applied to finned tube for increasing the heat transfer surface and compact design of construction area and the process fluid can be cooled by the natural air it is not required a coolant

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web our large fan field erected units are available in bay sizes up to 70 feet by 54 feet and fan diameters up to 36 feet our modular design and construction expertise allows for fast and dependable field erection benefits of the large fan design include air cooled heat exchanger air heat exchanger heat exchanger

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web the air cooler bank essentially consists of finned tube bundles fan rings axial fans with suitable drive units and a supporting structure it can be equipped individually with maintenance platforms louvers and other equipment as requested by the customer

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