

1. What is the term for metabolic pathways that release stored energy by breaking down complex molecules?

A. Anabolic pathways
 B. Catabolic pathways
 C. Fermentation process
 D. Thermodynamic pathways
 E. Bio energetic pathways
2. The molecule that functions as the reducing agent (electron donor) in a redox or oxidation-reduction reaction

A. Gains electron and gains energy
 B. Loses electrons and loses energy
 C. Gains electrons and loses energy
 D. Loses electrons and gains energy
 E. Neither gain nor loses electrons, but gains or loses energy
3. When electrons move closer to a more electronegative atom, what happens?

A. Energy is released
 B. Energy is consumed
 C. The more electronegative atom is reduced
 D. The more electronegative atom is oxidized
 E. A and C are correct
4. Why does the oxidation of organic compounds by molecular oxygen to produce CO₂ and water release free energy?

A. The covalent bond in organic molecules are higher energy bonds than those in water and carbon dioxide
 B. Electrons are being moved from atoms that have a lower affinity for electrons (such as C) to atoms with a higher affinity for electrons (such as O)
 C. The oxidation of organic compounds can be used to make ATP
 D. the electrons have a higher potential energy when associated with water and CO₂ than they do in organic compounds
 E. The covalent bonds in O₂ is unstable and easily broken by electron from organic molecules
5. Which of the following statements describes the results of this reaction?
 $C_6H_{12}O_6 + 6O_2 \rightarrow 6CO_2 + 6H_2O + \text{energy}$

A. C₆H₁₂O₆ is oxidized and O₂ is reduced
 B. O₂ is oxidized and H₂O is reduced
 C. CO₂ is reduced and O₂ is oxidized
 D. C₆H₁₂O₆ is reduced and CO₂ is oxidized
 E. O₂ is reduced and CO₂ is oxidized
6. When a glucose molecule loses a hydrogen atom as a result of oxidation-reduction reaction, the molecule becomes

A. Dehydrogenated
 B. Hydrogenated
 C. Oxidized
 D. Reduced
 E. An oxidizing agent
7. When a molecule of NAD⁺ (nicotinamide adenine dinucleotide) gains a hydrogen atom (not a hydrogen ion) the molecule becomes

A. Hydrogenated
 B. Oxidized
 C. Reduced
 D. Redoxed
 E. A reducing agent
8. Which of the following statements describes NAD⁺?

A. NAD⁺ is reduced to NADH during both glycolysis and the citric acid cycle
 B. NAD⁺ has more chemical energy than NADH
 C. NAD⁺ is reduced by the action of hydrogenases
 D. NAD⁺ can donate electrons for use in oxidative phosphorylation
 E. In the absence of NAD⁺, glycolysis can still function
9. Where does glycolysis take place?

A. Mitochondrial matrix
 B. Mitochondrial outer membrane
 C. Mitochondrial inner membrane
 D. Mitochondrial intermembrane space
 E. Cytosol
10. The ATP made during glycolysis is generated by

A. Substrate-level phosphorylation
 B. Electron transport
 C. Photo phosphorylation
 D. Chemiosmosis
 E. Oxidation of NAD⁺ and NADH
11. The oxygen consumed during cellular respiration is involved directly in which process or event?

A. Glycolysis
 B. Accepting electrons at the end of the electron transport chain
 C. The citric acid cycle
 D. The oxidation of pyruvate at acetyl CoA
 E. The phosphorylation of ADP to form ATP
12. Which process in eukaryote cells will proceed normally whether oxygen (O₂) is present or absent?

A. Electron transport
 B. Glycolysis
 C. The citric acid cycle
 D. Oxidation phosphorylation
 E. Chemiosmosis

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G Thomas



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Table of Contents Ap Biology Chapter 9 Reading Answers

1. Understanding the eBook Ap Biology Chapter 9 Reading Answers
 - The Rise of Digital Reading Ap Biology Chapter 9 Reading Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Ap Biology Chapter 9 Reading Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Ap Biology Chapter 9 Reading Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Ap Biology Chapter 9 Reading Answers
 - Personalized Recommendations
 - Ap Biology Chapter 9 Reading Answers User Reviews and Ratings
 - Ap Biology Chapter 9 Reading Answers and Bestseller Lists
5. Accessing Ap Biology Chapter 9 Reading Answers Free and Paid eBooks
 - Ap Biology Chapter 9 Reading Answers Public Domain eBooks
 - Ap Biology Chapter 9 Reading Answers eBook Subscription Services
 - Ap Biology Chapter 9 Reading Answers Budget-Friendly Options
6. Navigating Ap Biology Chapter 9 Reading Answers eBook Formats

- ePub, PDF, MOBI, and More
 - Ap Biology Chapter 9 Reading Answers Compatibility with Devices
 - Ap Biology Chapter 9 Reading Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Ap Biology Chapter 9 Reading Answers
 - Highlighting and Note-Taking Ap Biology Chapter 9 Reading Answers
 - Interactive Elements Ap Biology Chapter 9 Reading Answers
8. Staying Engaged with Ap Biology Chapter 9 Reading Answers
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Ap Biology Chapter 9 Reading Answers
9. Balancing eBooks and Physical Books Ap Biology Chapter 9 Reading Answers
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Ap Biology Chapter 9 Reading Answers
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Ap Biology Chapter 9 Reading Answers
- Setting Reading Goals Ap Biology Chapter 9 Reading Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Ap Biology Chapter 9 Reading Answers
- Fact-Checking eBook Content of Ap Biology Chapter 9 Reading Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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