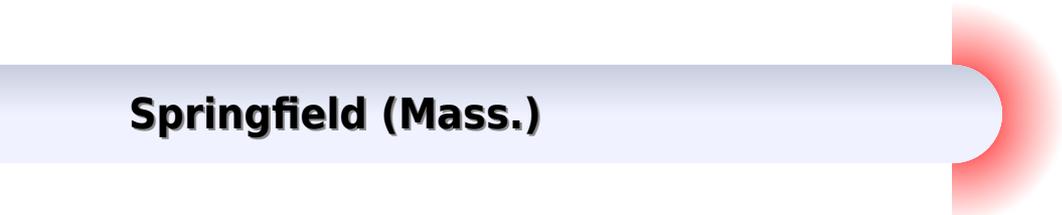


Tax rate	Single	Head of household	Married filing jointly	Married filing separately
10%	\$0 to \$10,275	\$0 to \$14,650	\$0 to \$20,550	\$0 to \$10,275
12%	\$10,276 to \$41,775	\$14,651 to \$55,900	\$20,551 to \$83,550	\$10,276 to \$41,775
22%	\$41,776 to \$89,075	\$55,901 to \$89,050	\$83,551 to \$178,150	\$41,776 to \$89,075
24%	\$89,076 to \$170,050	\$89,051 to \$170,050	\$178,151 to \$340,100	\$89,076 to \$170,050
32%	\$170,051 to \$215,950	\$170,051 to \$215,950	\$340,101 to \$431,900	\$170,051 to \$215,950
35%	\$215,951 to \$539,900	\$215,951 to \$539,900	\$431,901 to \$647,850	\$215,951 to \$323,925
37%	\$539,901 or more	\$539,901 or more	\$647,851 or more	\$323,926 or more

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Code of Federal Regulations, 1985 Hearings United States. Congress. House. Committee on Ways and Means, 1969

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web solution $v_1 = 2.50 \text{ l}$ $v_2 = t_1 \cdot 273 / 303 \text{ k}$ $t_2 = 273 / 70 \cdot 343 \text{ k}$ we will be applying charles law as pressure is constant here $v_1 / t_1 = v_2 / t_2$ or $v_2 = v_1 \cdot t_2 / t_1 = 2.50 \text{ l} \cdot 303 \text{ k} / 343 \text{ k} = 2.83 \text{ l}$ at 90 c a helium sample has a volume of 500 ml determine the temperature at which the volume of the liquid will become 240 ml