

CBSE Class 11 Chemistry Sample Paper Set 2

Time allowed: 3 Hours

Max. Marks: 70

Section-A (1 mark each)

General Instructions:

- All questions are compulsory
- Section A: Q.no. 1 to 5 are very short questions and carry 1 mark each.
- Section B: Q. no. 6 to 12 are short answer questions and carry 2 marks each.
- Section C: Q. no. 13 to 24 are also short answer questions and carry 3 marks each.
- Section D: Q. no. 25 to 27 are long answer questions and carry 5 marks each.
- There is no overall choice. However an internal choice has been provided in two questions of one mark, two questions of two marks, four questions of three marks and all the three questions of five marks weightage. You have to attempt only one of the choices in such questions.
- Use of log tables if necessary, use of calculators is not allowed.

Section-A (1 mark each)

Q1. What is the mass of a proton in amu?

OR

What is S.I unit of luminous intensity?

Q2. Bohrium has atomic no 107. Write its IUPAC name?

Q3. Mention the geometry of the following molecule?

OR

Name the region of electromagnetic spectrum to which the Balmer series of hydrogen belongs to?

Q4. Which method can be used to remove both temporary and permanent hardness of water?

Cbse Class 11 Chemistry Sample Paper 2014

Ying Liu



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