



Name:

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Making Salts Worksheet

This worksheet accompanies *Making Salts Summary* and *Precipitation Reactions Summary*.

1. When an acid reacts with a base, a neutralisation reaction occurs, producing a salt. Using the following equation, name the salts that are produced in the acid-base reactions.



- a) hydrochloric acid + ammonia →
- b) nitric acid + sodium hydroxide →
- c) phosphoric acid + potassium hydroxide →



2. There are three other ways of making salts from acids. Complete the following equations to illustrate them.

- a) acid + metal → salt +
- b) acid + → salt + water
- c) acid + carbonate → salt + + water

3. Use the above information to complete the following reactions that form soluble salts using acids.

- a) iron oxide + sulfuric acid → + water
- b) + hydrochloric acid → magnesium chloride + H₂O + CO₂
- c) zinc + → zinc nitrate + hydrogen

4. Name two ways in which salt is used by humans in everyday life.

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