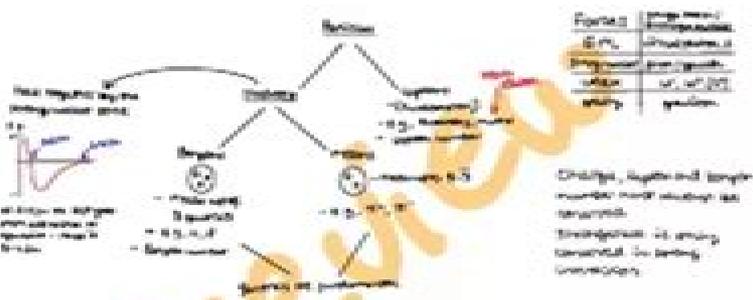


Particles and Quantum



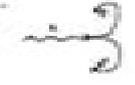
Force	Direction
EM	Electromagnetic
Gravitational	Gravitational
Weak	Weak
Strong	Strong

Charge, spin and baryon number must always be conserved. Conservation is only conserved in every interaction.

To convert mass to energy, use $E=mc^2$. Annihilation - pair of particles and its corresponding antiparticle collide and their rest energy is converted into photons.



Pair production - when a photon turns into a particle and its corresponding antiparticle. The photon must have at least the same amount of energy as the rest energy of the particles.



Excitation

Exciting means they gain energy through energy to climb their atomic shells. Alpha particles - Helium nucleus, ${}^4_2\text{He}$. - Highly ionising, so it is dangerous. - Relatively penetrating - stopped by a piece of paper or foil on skin. Beta - fast moving electron $\rightarrow e^-$. - Medium penetrating and relatively ionising - stopped by a few cm of aluminium. - Not ionising - ${}^0_{-1}\text{e}$ or ${}^0_{+1}\text{e}$ - penetrates to several centimeters.



Gamma - high energy. - EM ray or photon emitted from a high energy nucleus. - Highly penetrating - stopped by a few m of concrete.

Isotopes - Same element - same number of protons. - Different relative atomic mass - different number of neutrons. ${}^A_Z\text{X}$ or ${}^A_Z\text{C}$

Nuclear charge - $\frac{Q}{m}$ (C/kg) - charge to mass ratio.

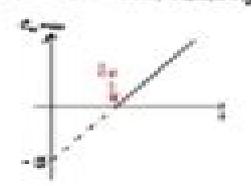
Radioactive - change of a nucleus when spontaneous through a bit of time. ${}^{238}_{92}\text{U}$ - ${}^{234}_{90}\text{Th}$ + ${}^4_2\text{He}$



Photoelectric effect

Use photons of sufficient energy on electrons on the surface of the metal, liberating them.

- 1. Photon absorbed - $E_{photon} \geq \phi$ (work function) - energy for electron to escape.
- 2. Frequency
- 3. Work function - minimum energy to liberate electrons.
- 4. Kinetic energy of electron after it is liberated.
- 5. Stopped potential - minimum potential needed for electrons to be liberated.



The photoelectric effect proved the particle nature of light due to the fact that the maximum kinetic energy of electrons is 0 if the frequency is less than the work function. Increasing the intensity of light would increase the number of electrons liberated but it would not affect the maximum kinetic energy of the electrons per photon.

Stopping potential - the pot. needed to stop any electrons from reaching the other side.

- 1. $E_{k,max} = hf - \phi$
- 2. Change in momentum
- 3. Stopping potential

De Broglie

The de Broglie wavelength is the wavelength that a particle can have. The wave nature of a particle was proven by diffraction of a crystal. This is because the electron is diffracting which is wave-like.



$$\lambda = \frac{h}{p} = \frac{h}{mv}$$

λ = de Broglie wavelength
 p = momentum / mass x velocity

Fluorescent tubes

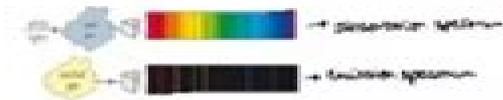


These tubes contain mercury under a high voltage is applied. This excites the gas resulting in which some of the mercury atoms emit photons when they relax. When the gas relaxes it emits UV light which is higher energy than what the electrons have, they emit photons in the UV range. A phosphor coating in the tube absorbs the photons and emit visible light.

Energy levels

Electrons can be excited to higher energy levels. Either by absorbing photons from the light or by the collision. In a gas discharge tube or in a gas discharge cell when ionizing the electron energy.

Ionization energy - is enough energy to pull an electron out of an atom, so will completely change the atom.



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