

## FORMULAS FOR IONIC COMPOUNDS

1. Write the chemical formula for the ionic compounds made from the following elements.

| Elements              | metal ion | non-metal ion | Chemical Formula | Name              |
|-----------------------|-----------|---------------|------------------|-------------------|
| potassium and iodine  | $K^{+1}$  | $I^{-1}$      | KI               | Potassium Iodide  |
| barium and chlorine   | $Ba^{2+}$ | $Cl^{-1}$     | $BaCl_2$         | Barium Chloride   |
| sodium and oxygen     | $Na^{+1}$ | $O^{2-}$      | $Na_2O$          | sodium Oxide      |
| aluminum and nitrogen | $Al^{3+}$ | $N^{3-}$      | AlN              | Aluminum Nitride  |
| lithium and sulfur    | $Li^{+1}$ | $S^{2-}$      | $Li_2S$          | Lithium Sulfide   |
| magnesium and oxygen  | $Mg^{+2}$ | $O^{2-}$      | MgO              | Magnesium Oxide   |
| calcium and chlorine  | $Ca^{2+}$ | $Cl^{-1}$     | $CaCl_2$         | calcium chloride  |
| boron and chlorine    | $B^{3+}$  | $Cl^{-1}$     | $BCl_3$          | boron chloride    |
| beryllium and sulfur  | $Be^{2+}$ | $S^{2-}$      | BeS              | beryllium sulfide |

2. Complete the chart below and write the ionic formulas to complete it. A few have been done for you.

|           | F       | Cl       | $O^{2-}$          | $S^{2-}$  | $N^{3-}$  | $P^{3-}$  |
|-----------|---------|----------|-------------------|-----------|-----------|-----------|
| $Li^{+}$  | LiF     | LiCl     | $Li_2O$           | $Li_2S$   | $Li_3N$   | $Li_3P$   |
| $Na^{+}$  | NaF     | NaCl     | Na <sub>2</sub> O | $Na_2S$   | $Na_3N$   | $Na_3P$   |
| $K^{+}$   | KF      | KCl      | $K_2O$            | $K_2S$    | $K_3N$    | $K_3P$    |
| $Be^{2+}$ | $BeF_2$ | $BeCl_2$ | BeO               | BeS       | $Be_3S_2$ | $Be_3P_2$ |
| $Mg^{2+}$ | $MgF_2$ | $MgCl_2$ | MgO               | MgS       | $Mg_3N_2$ | $Mg_3P_2$ |
| $Ca^{2+}$ | $CaF_2$ | $CaCl_2$ | CaO               | CaS       | $Ca_3N_2$ | $Ca_3P_2$ |
| $Al^{3+}$ | $AlF_3$ | $AlCl_3$ | $Al_2O_3$         | $Al_2S_3$ | AlN       | AlP       |
| $Ga^{3+}$ | $GaF_3$ | $GaCl_3$ | $Ga_2O_3$         | $Ga_2S_3$ | $GaN$     | GaP       |
| $B^{3+}$  | $BF_3$  | $BCl_3$  | $B_2O_3$          | $B_2S_3$  | BN        | BP        |

# Ionic Formula Writing Kit Answers

**Puja Mehta**



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**An Introduction to Chemistry** Michael Mosher, Paul Kelter, 2023-03-18 This textbook is written to thoroughly cover the topic of introductory chemistry in detail with specific references to examples of topics in common or everyday life It provides a major overview of topics typically found in first year chemistry courses in the USA The textbook is written in a conversational question based format with a well defined problem solving strategy and presented in a way to encourage readers to think like a chemist and to think outside of the box Numerous examples are presented in every chapter to aid students and provide helpful self learning tools The topics are arranged throughout the textbook in a traditional approach to the subject with the primary audience being undergraduate students and advanced high school students of chemistry

*Questions set at the examinations* College Entrance Examination Board, 1923 *Questions Set at the Examinations*, 1922 **Questions Set at the Comprehensive Examinations of ...**, 1924 *Chemical formulas and names* Virginia P. Powell, 1965 *Chemistry* James E. Brady, John R. Holum, 1995-12-29 Offers accurate lucid and interesting explanations of basic concepts and facts of chemistry while helping readers develop skills in analytical thinking and problems solving

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introduction antihistamines are medications that block the body from releasing histamines which is a substance that produces an inflammatory response many people use over the counter or prescription antihistamines to treat various allergies such as allergic rhinitis hay fever and other seasonal allergies etc

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