

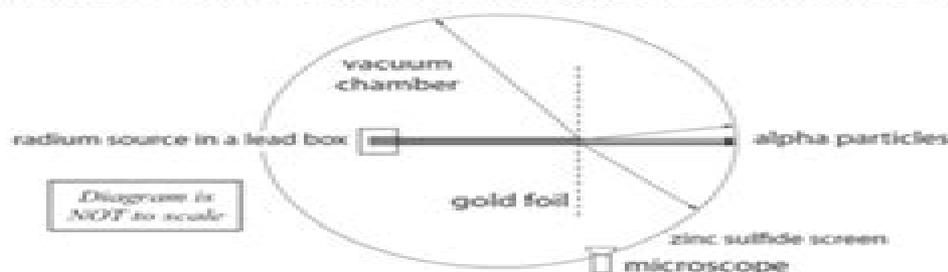
ATOMS: ATOMIC STRUCTURE QUESTIONS AND ANSWERS

QUESTION ONE: MODELS OF THE ATOM (2011:1)

At different times scientists have proposed various descriptions or models of the atom to match experimental evidence available.

- (a) The model that Thomson proposed was called the plum-pudding model. Describe this model.
Thomson's model had a number of negative particles / corpuscles / electrons embedded in a (solid) positive sphere / cloud.
Cloud of particles or energy not accepted.
The atom as a whole is neutral because the charges are balanced.
Term "nucleus" not accepted.

Geiger and Marsden performed a series of experiments under the direction of Ernest Rutherford which led to a new model of the atom. A model of the gold foil experiment is shown below.



- (b) For each of the conclusions given below, state which observation from the experiment provides evidence that:
- (i) Most of the mass of the atom is concentrated in a tiny region which Rutherford called the nucleus.
Most of the alpha particles went straight through and a very small number / a few bounced back / were reflected.
- (ii) The nucleus is positively charged.
Some of the alpha particles were deflected ("deflection" is an observation, "repulsion" is not).
- (c) Rutherford included several features in the experiment due to the characteristics of the alpha particles. Discuss the reason for each of the features below:
- (i) The source of the alpha particles is at the end of a narrow lead tube
Tube: To provide a narrow (or collimated) beam of alpha particle.
OR to direct the alpha particles
OR ensure alpha hits perpendicular to gold foil
OR to avoid stray alpha particles affecting the results
- (ii) Vacuum chamber
Vacuum chamber – Alpha particles would not travel far / be stopped / lose energy / path would change / be deviated / collide with the air particles (cannot accept "affect", "disrupt", "react", "interfere").
OR
Alpha would become a helium atom / alpha particles ionise the air.
- (iii) Zinc sulfide screen
Alpha particles are not visible to the naked eye. The ZnS coating will show / glow / fluoresce / flash / give off light when alpha particles strike it (this may be implied as opposed to directly stated) but do not allow "detected" or "located"

Atoms Atomic Structure Questions And Answers

Albert Cushing Crehore



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