

## Chemical Analysis – What can you Remember?

1. What colour precipitate do chloride ions produce?
  - a) White
  - b) Cream
  - c) Yellow
  - d) Green
2. What colour do sodium ions go in a flame test?
  - a) Red
  - b) Yellow
  - c) Brick-red
  - d) Lilac
3. Which two solutions are added to test for the presence of halide ions?
  - a) Silver chloride and nitric acid
  - b) Barium chloride and hydrochloric acid
  - c) Silver nitrate and nitric acid
  - d) Barium chloride and sulphuric acid
4. What is the test for oxygen gas?
  - a) Lit splint goes out with a squeaky pop
  - b) Damp blue litmus paper goes red
  - c) Damp red litmus paper goes blue
  - d) Glowing splint relights
5. What colour precipitate do iodide ions form?
  - a) White
  - b) Cream
  - c) Yellow
  - d) Green
6. Which ion produces a brick-red colour in a flame test?
  - a) Sodium
  - b) Calcium
  - c) Copper
  - d) Lithium
7. What is the test for carbon dioxide gas?
  - a) Limewater turns cloudy
  - b) Damp blue litmus turns red
  - c) A glowing splint relights
  - d) A lit splint goes out with a squeaky pop
8. When sodium hydroxide is added to a solution containing iron (III) ions, what colour precipitate forms?
  - a) White
  - b) Blue
  - c) Green
  - d) Red/brown
9. What is the test for chlorine gas?
  - a) Damp blue litmus paper turns red and is then bleached white
  - b) Red litmus  $\square$  turns blue
  - c) A glowing splint relights
  - d) Limewater will turn cloudy
10. Copper (II) ions form what colour precipitate test?
  - a) Blue
  - b) Green
  - c) Red/brown
  - d) Cream
11. Which two solutions are added to test for the presence of sulphate ions?
  - a) Silver chloride and nitric acid
  - b) Barium chloride and sulphuric acid
  - c) Silver nitrate and nitric acid
  - d) Barium chloride and hydrochloric acid
12. What colour do potassium ions go in a flame test?
  - a) Lilac
  - b) Red
  - c) Yellow
  - d) Brick-red
13. What is the test for hydrogen gas?
  - a) Lit splint goes out with a squeaky pop
  - b) Glowing splint relights
  - c) Damp blue litmus paper turns red and then is bleached white
  - d) Limewater goes cloudy
14. Which ion produces a red colour in a flame test?
  - a) Sodium
  - b) Calcium
  - c) Copper
  - d) Lithium
15. What colour precipitate do bromide ions form?
  - a) White
  - b) Cream
  - c) Yellow
  - d) Green
16. What needs to be added to test for the presence of carbonate ions?
  - a) Silver nitrate
  - b) Universal indicator
  - c) Litmus paper
  - d) Dilute acid
17. If sodium hydroxide is added to a compound containing ammonium ions, what gas is produced?
  - a) Oxygen
  - b) Ammonia
  - c) Carbon dioxide
  - d) Hydrogen
18. What colour precipitate does barium sulphate form?
  - a) White
  - b) Green
  - c) Yellow
  - d) Cream
19. What is needed to carry out a flame test?
  - a) Splint
  - b) Glass rod
  - c) Spatula
  - d) Nichrome/platinum wire

Your score out of 19=  
Percentage =

# Chemistry Unit 2 Test Answers

**KJ Lindholm-Leary**



**Chemistry Unit 2 Test Answers:**

## Whispering the Techniques of Language: An Emotional Quest through **Chemistry Unit 2 Test Answers**

In a digitally-driven world wherever screens reign supreme and immediate connection drowns out the subtleties of language, the profound techniques and mental subtleties concealed within phrases usually get unheard. However, nestled within the pages of **Chemistry Unit 2 Test Answers** a interesting literary treasure pulsating with raw feelings, lies a fantastic journey waiting to be undertaken. Penned by a talented wordsmith, that marvelous opus invites readers on an introspective journey, lightly unraveling the veiled truths and profound impact resonating within the very fabric of every word. Within the emotional depths with this emotional review, we shall embark upon a genuine exploration of the book is core subjects, dissect their charming publishing fashion, and fail to the effective resonance it evokes heavy within the recesses of readers hearts.

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