

# 11 • Chemistry of Carbon

- A molecule named propene has the formula:
  - $C_3H_6$
  - $C_2H_2$
  - $C_4H_8$
  - $C_3H_8$
  - $C_3H_4$

*Carbon*
- What is the name of a molecule with the formula,  $C_5H_{10}$ ?
  - pentene
  - pentane
  - butyne
  - butene

*Carbon*
- A molecule that fits the pattern,  $C_nH_{2n+2}$ , is a(n)
  - alkane
  - alkene
  - cyclic compound
  - alkyne
  - need more info
- The general formula of the alkyne series is:
  - $C_nH_{2n}$
  - $C_nH_{2n-4}$
  - $C_nH_{2n-2}$
  - $C_nH_{2n-6}$
- How many  $\pi$  bonds are present in propyne?
  - 2
  - 1
  - 3
  - 4

*C-C≡C*
- The hydrocarbon series that has a double bond between carbon atoms is the
  - alkene family
  - alkane family
  - alkyne family
  - alkadiene family
- Organic chemistry is the study of
  - compounds of carbon
  - nonmetals
  - once-living things
  - compounds produced by living things

*(HISTORIC IDEA)*

## PRACTICE TEST

- In octene, the hybridization(s) of carbon is(are):
  - $sp^3$  and  $sp^2$
  - $sp^2$  only
  - $sp^3$  only
  - $sp$ ,  $sp^2$  and  $sp^3$

*-C-C=C-C-C-C-C-*
- The properties of carboxylic acids include all of the following except:
  - they can be neutralized by bases *TRUE*
  - the molecule has some polarity *TRUE*
  - they ionize completely in water solution
  - they react with alcohols to form esters *TRUE*
- Dimethylbenzene is also known as xylene. 1,3-dimethylbenzene would be the same as...
  - meta-xylene
  - ortho-xylene
  - para-xylene
  - none of these
- The compound
 

is classified as a

  - ketone
  - alcohol
  - ester
  - aldehyde
  - acid
- What hybridization is utilized by the oxygen atom in the carbonyl group above?
  - $sp^2$
  - $sp^3$
  - $sp$

*-C=O: SN=3  
sp<sup>2</sup>*

# Chemistry Chapter 13 Test Answer Key

**Alan Sherman**



## **Chemistry Chapter 13 Test Answer Key:**

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## **Table of Contents Chemistry Chapter 13 Test Answer Key**

1. Understanding the eBook Chemistry Chapter 13 Test Answer Key
  - The Rise of Digital Reading Chemistry Chapter 13 Test Answer Key
  - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Chapter 13 Test Answer Key
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in a Chemistry Chapter 13 Test Answer Key
  - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Chapter 13 Test Answer Key
  - Personalized Recommendations
  - Chemistry Chapter 13 Test Answer Key User Reviews and Ratings

- Chemistry Chapter 13 Test Answer Key and Bestseller Lists
- 5. Accessing Chemistry Chapter 13 Test Answer Key Free and Paid eBooks
  - Chemistry Chapter 13 Test Answer Key Public Domain eBooks
  - Chemistry Chapter 13 Test Answer Key eBook Subscription Services
  - Chemistry Chapter 13 Test Answer Key Budget-Friendly Options
- 6. Navigating Chemistry Chapter 13 Test Answer Key eBook Formats
  - ePub, PDF, MOBI, and More
  - Chemistry Chapter 13 Test Answer Key Compatibility with Devices
  - Chemistry Chapter 13 Test Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Chemistry Chapter 13 Test Answer Key
  - Highlighting and Note-Taking Chemistry Chapter 13 Test Answer Key
  - Interactive Elements Chemistry Chapter 13 Test Answer Key
- 8. Staying Engaged with Chemistry Chapter 13 Test Answer Key
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Chemistry Chapter 13 Test Answer Key
- 9. Balancing eBooks and Physical Books Chemistry Chapter 13 Test Answer Key
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Chemistry Chapter 13 Test Answer Key
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Chemistry Chapter 13 Test Answer Key
  - Setting Reading Goals Chemistry Chapter 13 Test Answer Key
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chemistry Chapter 13 Test Answer Key
  - Fact-Checking eBook Content of Chemistry Chapter 13 Test Answer Key
  - Distinguishing Credible Sources

13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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