

11-3 Practice Problems

limiting reactants - part 2

Name: _____

Date: _____

1. If 4.1 g of Cr is heated with 9.3 g of Cl_2 , what mass of CrCl_3 will be produced?



$$4.1\text{g} \times \frac{1\text{mole}}{52\text{g}} = 0.078\text{moles Cr}$$

$$9.3\text{g} \times \frac{1\text{mole}}{70\text{g}} = 0.133\text{moles Cl}_2$$

$$0.078\text{moles} \times \frac{3}{2} = 0.1182\text{moles Cl}_2 \text{ needed. Do I have it?}$$

2. What mass of SO_2 is produced from the reaction between 31.5 g of S_8 and 8.65 g of O_2 ?



3. What mass of SO_2 is produced from the reaction of 12.4 g of SO_2 and 3.45 g of O_2 ?



4. What mass of SO_2 is produced from the reaction of 58 g of SO_2 and 1.64 g of H_2O ?



11 3 Practice Answers

Shasha Hu



11 3 Practice Answers:

Unveiling the Energy of Verbal Artistry: An Mental Sojourn through **11 3 Practice Answers**

In a global inundated with screens and the cacophony of immediate connection, the profound energy and mental resonance of verbal artistry usually fade in to obscurity, eclipsed by the continuous assault of sound and distractions. However, situated within the lyrical pages of **11 3 Practice Answers**, a charming perform of literary splendor that impulses with natural emotions, lies an unique trip waiting to be embarked upon. Published by a virtuoso wordsmith, that mesmerizing opus instructions readers on a mental odyssey, gently revealing the latent potential and profound affect stuck within the intricate internet of language. Within the heart-wrenching expanse of this evocative examination, we will embark upon an introspective exploration of the book is main themes, dissect their fascinating publishing type, and immerse ourselves in the indelible impact it leaves upon the depths of readers souls.

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