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**PT Brinkman**



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web a nucleus changes into a new element by emitting nuclear radiations these changes are described using nuclear equations example  ${}_{86}^{219}\text{textrm{rn}} \rightarrow {}_{84}^{215}\text{textrm{po}}$

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top number  $A$  represents the nucleon number or the mass number nucleon number a total number of protons and neutrons in the nucleus the lower number  $Z$  represents the proton or atomic number

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web sep 12 2022 the half life of the nuclei and thus the material is  $t_{1/2} = 693$  million years 35 a the decay constant is  $\lambda = 1.99 \times 10^{-5} \text{ s}^{-1}$  b since strontium 91 has an atomic mass of 90.90 g

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