

Chapter 9 Practice Problems: Chemical Names and Formulas

Section 9.1: Naming Ions

- What is the charge on the ion typically formed by each element?
 - oxygen
 - iodine
 - sodium
 - aluminum
 - nickel, two electrons lost
 - magnesium
- How many electrons does the neutral atom gain or lose when each ion forms?
 - Cr^{3+}
 - P^{3-}
 - Li^+
 - Ca^{2+}
 - Cl^-
 - O^{2-}
- Name each ion. Identify each as a cation or an anion.
 - Sn^{2+}
 - Co^{3+}
 - Br^-
 - K^+
 - H^-
 - Mn^{2+}
- Write the formula (including the charge) for each ion. Use table 9.3 as if necessary.
 - carbonate ion
 - nitrite ion
 - sulfate ion
 - hydroxide ion
 - chromate ion
 - ammonium ion
- Name each of the following ions. Identify each as a cation or an anion.
 - CN^-
 - HCO_3^-
 - PO_4^{3-}
 - Cl^-
 - Ca^{2+}
 - SO_3^{2-}

9.2 Naming and Writing Formulas for Ionic Compounds:

- Write the formulas for these binary ionic compounds.
 - magnesium oxide
 - tin(II) fluoride
 - potassium iodide
 - aluminum chloride
 - sodium sulfide
 - ferric bromide
- Write the formulas for the compounds formed from these pairs of ions.
 - Ba^{2+} and Cl^-
 - Ag^+ and I^-
 - Ca^{2+} and S^{2-}
 - K^+ and Br^-
 - Al^{3+} and O^{2-}
 - Fe^{2+} and O^{2-}
- Name the following binary ionic compounds.
 - MnO_2
 - Li_3N
 - CaCl_2
 - SrBr_2
 - NiCl_2
 - K_2S
 - Fe_3N_2
 - SnCl_4
- Write formulas for the following ionic compounds.
 - sodium phosphate
 - magnesium sulfate
 - sodium hydroxide
 - potassium cyanide
 - ammonium chloride
 - potassium dichromate
- Write formulas for the following ionic compounds.
 - NH_4^+ and SO_4^{2-}
 - K^+ and NO_3^-
 - barium ion and hydroxide ion
 - lithium ion and carbonate ion

Chapter 9 Chemical Names Formulas Practice Problems Answers

CO Houle



Chapter 9 Chemical Names Formulas Practice Problems Answers:

The Top Books of the Year Chapter 9 Chemical Names Formulas Practice Problems Answers The year 2023 has witnessed a remarkable surge in literary brilliance, with numerous compelling novels captivating the hearts of readers worldwide. Lets delve into the realm of bestselling books, exploring the fascinating narratives that have enthralled audiences this year.

Chapter 9 Chemical Names Formulas Practice Problems Answers : Colleen Hoover's "It Ends with Us" This heartfelt tale of love, loss, and resilience has captivated readers with its raw and emotional exploration of domestic abuse. Hoover masterfully weaves a story of hope and healing, reminding us that even in the darkest of times, the human spirit can succeed.

Chapter 9 Chemical Names Formulas Practice Problems Answers : Taylor Jenkins Reid's "The Seven Husbands of Evelyn Hugo" This intriguing historical fiction novel unravels the life of Evelyn Hugo, a Hollywood icon who defies expectations and societal norms to pursue her dreams. Reid's compelling storytelling and compelling characters transport readers to a bygone era, immersing them in a world of glamour, ambition, and self-discovery.

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The novel begins with Richard Papen, a bright but troubled young man, arriving at Hampden College. Richard is immediately drawn to the group of students who call themselves the Classics Club. The club is led by Henry Winter, a brilliant and charismatic young man. Henry is obsessed with Greek mythology and philosophy, and he quickly draws Richard into his world. The other members of the Classics Club are equally as fascinating. Bunny Corcoran is a wealthy and spoiled young man who is always looking for a good time. Charles Tavis is a quiet and reserved young man who is deeply in love with Henry. Camilla Macaulay is a beautiful and intelligent young woman who is drawn to the power and danger of the Classics Club. The students are all deeply in love with Morrow, and they are willing to do anything to please him. Morrow is a complex and mysterious figure, and he seems to be manipulating the students for his own purposes. As the students become more involved with Morrow, they begin to commit increasingly dangerous acts. The Secret History is a masterful and thrilling novel that will keep you guessing until the very end. The novel is a cautionary tale about the dangers of obsession and the power of evil.

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