

Chapter Test B

Chapter: Chemical Equations and Reactions

Match a name space (possibly writing the name of the react or product molecule) completed with reactants/products atoms/molecules.

- The product(s) of a chemical reaction will be represented by a double arrow pointing to the right and the reactant(s) will be represented by a double arrow pointing to the left.
 - reactants**
 - products
 - solvents
- An ionic chemical equation is represented by adding the
 - charges
 - charges of the particles
 - number of particles
- An ionic equation shows a reaction by adding particles that are in the
 - state
 - aqueous or solid
 - solvent
- The activity for precipitation reactions are used to predict whether
 - soluble ionic compounds will react to form an insoluble precipitate
 - insoluble ionic compounds will react to form a soluble precipitate
 - soluble ionic compounds will react to form a precipitate
 - insoluble ionic compounds will react to form a precipitate
- The coefficients are chemical equations
 - indicate the number of moles of each substance that react
 - show the number of grams of each substance that react
 - show the number of moles of the substances
 - show the relative proportions of each atom
- If you're to balance a chemical equation with one variable, then
 - X will replace one of the coefficients
 - X will replace one of the subscripts
 - X will replace one of the elements
 - X will replace both sides

Chapter 8 Test B

**David Greenwood, David
Robertson, Sarah Woolley, Jenny
Goodman, Jenny Vaughan**

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Mathematics for the Australian Curriculum Year 10 2ed Teacher Support Print Option David Greenwood, David Robertson, Sarah Woolley, Jenny Goodman, Jenny Vaughan, 2015-11-19

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