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Advanced
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convert to basic solution $4\text{CN} + 2\text{Au} + \text{CN} + 2\text{e}^-$ multiplied by a factor of 2 $2\text{e}^- + 2\text{H}_2\text{O} \rightarrow \text{O}_2 + 2\text{OH}^-$ the final answer

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every atom of nickel oxidized but where do those electrons go

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