

P_1 = Pressure of working substance entering the system

V_1 = Specific volume of working substance

U_1 = Specific Internal Energy

C_1 = Velocity of working substance

Z_1 = Height above the datum level

P_2, V_2, U_2, C_2, Z_2 = Corresponding values of working substance

Q = Heat supplied to system

W = Work developed by system

Total energy at Entering = P.E + K.E + U.E + F.E + Heat energy

$$= gz_1 + \frac{C_1^2}{2} + (U_1 + P_1 V_1) + Q$$

Total energy at Leaving = P.E + K.E + U.E + F.E + Work done

$$= gz_2 + \frac{C_2^2}{2} + (U_2 + P_2 V_2) + W$$

Total energy entering and leaving,

$$gz_1 + \frac{C_1^2}{2} + (U_1 + P_1 V_1) + Q = gz_2 + \frac{C_2^2}{2} + (U_2 + P_2 V_2) + W$$

We know that,

$$H = U + PV$$

$$gz_1 + \frac{C_1^2}{2} + H_1 + Q = gz_2 + \frac{C_2^2}{2} + H_2 + W$$

Multiply 'm' on both sides,

$$m [gz_1 + \frac{C_1^2}{2} + H_1 + Q] = m [gz_2 + \frac{C_2^2}{2} + H_2 + W]$$

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Camilla Rothe



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 - The Rise of Digital Reading Engineering Thermodynamics Problems And Solutions Bing
 - Advantages of eBooks Over Traditional Books
2. Identifying Engineering Thermodynamics Problems And Solutions Bing
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Engineering Thermodynamics Problems And Solutions Bing
 - User-Friendly Interface
4. Exploring eBook Recommendations from Engineering Thermodynamics Problems And Solutions Bing
 - Personalized Recommendations
 - Engineering Thermodynamics Problems And Solutions Bing User Reviews and Ratings
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 - Engineering Thermodynamics Problems And Solutions Bing Compatibility with Devices
 - Engineering Thermodynamics Problems And Solutions Bing Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Engineering Thermodynamics Problems And Solutions Bing
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 - Interactive Elements Engineering Thermodynamics Problems And Solutions Bing
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 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Engineering Thermodynamics Problems And Solutions Bing
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 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Engineering Thermodynamics Problems And Solutions Bing
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 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Engineering Thermodynamics Problems And Solutions Bing
 - Setting Reading Goals Engineering Thermodynamics Problems And Solutions Bing
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Engineering Thermodynamics Problems And Solutions Bing
 - Fact-Checking eBook Content of Engineering Thermodynamics Problems And Solutions Bing
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
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 - Integration of Multimedia Elements

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