

# Introduction to Chemistry

## Section 1.1 The Stories of Two Chemicals

In your textbook, read about the ozone layer.

The words of the boxes below just need to complete the passage.

atmosphere	oxygen gas	ozone	ozone hole
stratosphere	hydrophobic	ultraviolet radiation	

Earth's **OA** \_\_\_\_\_ is made up of several layers. The air molecules gather up near the lower level. This layer is called the **OB** \_\_\_\_\_. The next layer up is called the **OC** \_\_\_\_\_. This layer contains a protective \_\_\_\_\_.

**OD** \_\_\_\_\_ layer

Chlorine **OE** \_\_\_\_\_ is made by ultraviolet radiation in the upper part of the **OC** \_\_\_\_\_. The ozone **OF** \_\_\_\_\_ layer around Earth, which shields \_\_\_\_\_.

**OG** \_\_\_\_\_ Without ozone, you get more **OH** \_\_\_\_\_ to get a rash or possibly skin cancer. The thinning of the ozone layer, called the **OI** \_\_\_\_\_, is measured because without ozone all organisms on Earth are subject to more **OJ** \_\_\_\_\_.

In your textbook, read about chlorofluorocarbons.

The words statements below, rephrase as follows:

- \_\_\_\_\_ **I**. CFCs is another name for a chlorofluorocarbon.
- \_\_\_\_\_ **II**. CFCs are made up of carbon, fluorine, and chlorine.
- \_\_\_\_\_ **III**. All CFCs are synthetic chemicals.
- \_\_\_\_\_ **IV**. CFCs readily react easily with other chemicals.
- \_\_\_\_\_ **V**. CFCs were developed as replacements for toxic refrigerants.

# Chapter 11 Study For Content Mastery

**Wanda Webb, Richard K. Adler**



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